

QUESTION 1

- 1.1. C ✓✓
- 1.2. D ✓✓
- 1.3. B ✓✓
- 1.4. A ✓✓
- 1.5. A ✓✓

[10]

QUESTION 2

- 2.1.
 - 2.1.1. A pure substance is a substance that cannot be separated into simpler components by physical methods. ✓✓
 - 2.1.2.1 Element ✓
 - 2.1.2.2 Consists of only one type of atom. ✓
 - 2.1.2.3 It is a combination of different metals (copper and zinc). ✓
- 2.2. Pots and pans are made of metal, because metal is a good conductor which allows heat to be transferred so that food can cook. ✓ The handles are insulators so that you do not burn your hands when you pick up a hot pot. ✓
- 2.3.1.
 - (a) Temperature ✓
 - (b) Time ✓
- 2.3.2. Boiling point is the temperature of a liquid at which its vapour pressure equals to the external (atmospheric) pressure ✓✓
- 2.3.3. The temperature remains constant because the energy is absorbed by the molecules. ✓
The strength of the intermolecular forces between the molecules decreases. ✓
The particles move faster and spaces between the particles increases. ✓
The substance changes phase from liquid to gas ✓

[15]

Physical Sciences P2 2010 Memorandum June

Siddappa N.Byrareddy



Physical Sciences P2 2010 Memorandum June:

The book delves into Physical Sciences P2 2010 Memorandum June. Physical Sciences P2 2010 Memorandum June is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Sciences P2 2010 Memorandum June, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 1: Introduction to Physical Sciences P2 2010 Memorandum June
 - Chapter 2: Essential Elements of Physical Sciences P2 2010 Memorandum June
 - Chapter 3: Physical Sciences P2 2010 Memorandum June in Everyday Life
 - Chapter 4: Physical Sciences P2 2010 Memorandum June in Specific Contexts
 - Chapter 5: Conclusion
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