

## Density Worksheet

### Physical Science

$D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm <sup>3</sup> )		Substance	Density (g/cm <sup>3</sup> )
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

- The ratio of an object's mass to its \_\_\_\_\_ is called the *density* of the object.
- A kilogram of lead occupies a much smaller volume than a kilogram of water, because \_\_\_\_\_ has a much higher *density*.
- For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
  - mass = 453 g; volume = 225 cm<sup>3</sup>
  - mass = 5.0 g; volume = 10.0 cm<sup>3</sup>
  - mass = 26.1 g; volume = 2.0 mL
- If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
- A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

# Physical Science Worksheet Answers

**Jin-Ying Zhang**



## **Physical Science Worksheet Answers:**

## **Physical Science Worksheet Answers** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Physical Science Worksheet Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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