

QUESTION 1: MULTIPLE-CHOICE QUESTIONS

Various options are provided as possible answers to the following questions. Choose the answer and write only the letter (A-D) next to the question number (1.1 - 1.10) in the ANSWER BOOK, for example 1.5 D.

1.1 Which ONE of the following is a molecule with a multiple bond?

- A N_2
- B NH_3
- C OF_2
- D HOCl

(2)

1.2 The molecular formula of ethyne is C_2H_2 .

The shape of the ethyne molecule as predicted by the VSEPR theory is:

- A Trigonal planar.
- B Octahedral.
- C Linear.
- D Bent.

(2)

1.3 The table below indicates the boiling points of four liquids.

SUBSTANCE	BOILING POINT ($^{\circ}\text{C}$)
Water	100
Methylated spirits	78,5
Ethanol	78,37
Acetone	56

Which ONE of the above liquids has the lowest surface tension?

- A Water.
- B Methylated spirits.
- C Ethanol.
- D Acetone.

(2)

1.4 The predominant forces between the molecules in an ice crystal are called . . .

- A London forces.
- B Hydrogen bonding.
- C Induced dipole forces.
- D Polar covalent bonding.

(2)
[8]

Physical Science Past Exam Papers 2013

Christian G. Meyer



Physical Science Past Exam Papers 2013:

This book delves into Physical Science Past Exam Papers 2013. Physical Science Past Exam Papers 2013 is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Science Past Exam Papers 2013, encompassing both the fundamentals and more intricate discussions.

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