



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

Duration: 120
Time: 1 hour



The question paper consists of 12 pages, including 4 extra sheets.
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QUESTION 8

PHYSICAL SCIENCES PAPER 2

ACID AND BASES

GRADE 11

NOVEMBER 2020

QUESTION 8: ACID AND BASES (10)

8.1. Give a chemical equation (show the full path) for the reaction to represent B.

The following reaction is shown on the page:

a. Name of the reaction (2)
b. Chemical equation (10)

QUESTION 8 (Start on a NEW page)

8.2. Sodium carbonate is an ionic compound with the following percentage composition:

Na 27,0%, C 14,3%, O 58,7%

The molar mass of the compound is 106 g mol⁻¹.

8.2.1. Calculate the mass percentage of Na (2)
8.2.2. Calculate, by inspection, the empirical formula (2)
8.2.3. Calculate the relative formula mass (2)

8.3. Concentrated sulphuric acid is a mixture of sulphuric acid and water. It is a mixture of sulphuric acid and water. It is a mixture of sulphuric acid and water.

8.3.1. Calculate the concentration of the sulphuric acid (2)
8.3.2. Calculate the concentration of the sulphuric acid (2)

8.4. Sodium carbonate reacts with an acid to form a salt. The salt is sodium carbonate.

2Na₂CO₃ + 4HCl → 2NaCl + 2H₂O

Calculate the volume of hydrogen gas (H₂) that would be produced at STP if 10 g of sodium carbonate reacts completely. (10)

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Table of Contents Physical Science2exam Paper For Grade11

1. Understanding the eBook Physical Science2exam Paper For Grade11
 - The Rise of Digital Reading Physical Science2exam Paper For Grade11
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science2exam Paper For Grade11
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science2exam Paper For Grade11
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science2exam Paper For Grade11
 - Personalized Recommendations
 - Physical Science2exam Paper For Grade11 User Reviews and Ratings

- Physical Science2exam Paper For Grade11 and Bestseller Lists
- 5. Accessing Physical Science2exam Paper For Grade11 Free and Paid eBooks
 - Physical Science2exam Paper For Grade11 Public Domain eBooks
 - Physical Science2exam Paper For Grade11 eBook Subscription Services
 - Physical Science2exam Paper For Grade11 Budget-Friendly Options
- 6. Navigating Physical Science2exam Paper For Grade11 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science2exam Paper For Grade11 Compatibility with Devices
 - Physical Science2exam Paper For Grade11 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science2exam Paper For Grade11
 - Highlighting and Note-Taking Physical Science2exam Paper For Grade11
 - Interactive Elements Physical Science2exam Paper For Grade11
- 8. Staying Engaged with Physical Science2exam Paper For Grade11
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science2exam Paper For Grade11
- 9. Balancing eBooks and Physical Books Physical Science2exam Paper For Grade11
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science2exam Paper For Grade11
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science2exam Paper For Grade11
 - Setting Reading Goals Physical Science2exam Paper For Grade11
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science2exam Paper For Grade11
 - Fact-Checking eBook Content of Physical Science2exam Paper For Grade11
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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