



GRADE 10 PHYSICAL SCIENCES CONTROL TEST  
MEMO

**QUESTION 1**

- 1.1. B ✓✓  
1.2. B ✓✓  
1.3. A ✓✓  
1.4. A ✓✓  
1.5. B ✓✓  
1.6. C ✓✓  
1.7. B ✓✓  
1.8. A ✓✓

[16]

**QUESTION 2**

- 2.1.1 Starts from rest/ $0 \text{ m}\cdot\text{s}^{-1}$  ✓  
Velocity increases at a constant rate ✓  
until he reaches  $2,5 \text{ m}\cdot\text{s}^{-1}$  after 25 s. ✓

**OR**

- Starts from rest/ $0 \text{ m}\cdot\text{s}^{-1}$  ✓  
Constant positive acceleration ✓  
until he reaches  $2,5 \text{ m}\cdot\text{s}^{-1}$  after 25 s. ✓

(3)

- 2.1.2 Constant/uniform velocity ✓  
for another 25 s. ✓

**OR**

- Zero/No acceleration ✓  
for another 25 s. ✓

**OR**

- Velocity remains  $2,5 \text{ m}\cdot\text{s}^{-1}$  in the  
direction of motion ✓  
for another 25 s. ✓

(2)

- 2.2.1  $acceleration = \frac{\Delta v}{\Delta t}$  ✓  
 $= \frac{0 - 2,5}{100 - 80}$  ✓  
 $= -0,0625$

$a = 0,063 \text{ m}\cdot\text{s}^{-2}$  ✓ opposite to direction of motion ✓

(4)

- 2.2.2 Length of track = Area between the graph and the time axis ✓  
 $= \frac{1}{2} (2,5) (35 + 100)$  ✓  
 $= 168,75 \text{ m}$  ✓

**OR**

- Length of track = Area of trapezium ✓  
 $= \frac{1}{2} (2,5) (35 + 100)$  ✓  
 $= 168,75 \text{ m}$  ✓

**OR**

- Length of track = Area between the graph and the time axis  
 $= \frac{1}{2} bh + \frac{1}{2} bh + lb$   
 $= \frac{1}{2} (2,5) + \frac{1}{2} (40)(2,5) + (35)(2,5)$   
 $= 31,2 + 87,5 + 50$   
 $= 168,75 \text{ m}$  ✓

(4)  
[13]

# Physical Sciences Papergradecontrolled Test

**Annelies Wilder-Smith**



## **Physical Sciences Papergradecontrolled Test:**

**Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed ,2000      Resources in Education ,1998      Research in Education ,1974      Resources for Teaching Middle School Science** Smithsonian Institution,National Academy of Engineering,National Science Resources Center of the National Academy of Sciences,Institute of Medicine,1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents      **Annals of the American Academy of Political and Social Science ,1924      New Scientist ,1975      The Annals of the American Academy of Political and Social Science ,1924      Chemical News and Journal of Physical Science ,1888      Chemical News and Journal of Physical Science** William Crookes,James H. Gardiner,Gerald Druce,H. W. Blood-Ryan,1888      The Chemical News and

Journal of Physical Science ,1888      *A Textbook of Sports Science : TEST, EVALUATION, ACCREDITATION, MEASUREMENTS And STANDARDS ( TEAMS )* DR. DEVINDER K. KANSAL,2021-07-23 CONTENTS IN BRIEF PREFACE ACKNOWLEDGEMENTS vii PART I INTRODUCTION 19 77 Chapter 1 Philosophy of Testing 21 Chapter 2 Need for Sports Science to Develop Sports Excellence 36 Chapter 3 Measuring Physical Education Component is Lifeline of All Education 52 Chapter 4 History of Test and Measurement 68 PART II TEST CONSTRUCTION 78 143 Chapter 5 Test Classification 80 Chapter 6 Criteria of Good Test 88 Chapter 7 Construction of Psychomotor Tests 104 Chapter 8 Construction of Knowledge Tests 116 Chapter 9 Construction of Affective Tests 126 Chapter 10 Test Administration 131 PART III PHYSICAL TESTS 144 185 Chapter 11 Anthropometric Tests 145 Chapter 12 Testing Health Markers 164 PART IV PSYCHOMOTOR TESTS 186 373 Chapter 13 Introduction to Psychomotor Testing 188 Chapter 14 Muscular Strength Endurance Testing 198 Chapter 15 Cardio Pulmonary Endurance 212 Chapter 16 Flexibility Tests 231 Chapter 17 Muscular Power Tests 244 Chapter 18 Agility Tests 252 Chapter 19 Balance and Its Tests 260 Chapter 20 Tests of Speed Reaction Time and Coordination 272 Chapter 21 Physical Activity Cognition and Testing 284 Chapter 22 Physical and Motor Fitness Testing 298 Chapter 23 General Motor Ability Testing 307 Chapter 24 Team Games Skill Testing 317 Chapter 25 Individual Games Skill Testing 348 PART V MEASUREMENTS 374 418 Chapter 26 Measurements of Champions 375 Chapter 27 Measurement of Behaviour Change and Sport for All 385 Chapter 28 Measurement and Sports Talent Selection 397 Chapter 29 Measurement and Sports Excellence 411 PART VI STATISTICS EVALUATION ACCREDITATION STANDARDS 419 536 Chapter 30 Introduction to Statistical Tests 421 Chapter 31 Data Distribution and Central Tendency 429 Chapter 32 Variability Testing 446 Chapter 33 Normal Probability Curve 454 Chapter 34 Diagrammatic Representations of Data 458 Chapter 35 Evaluation Fundamentals 478 Chapter 36 Accreditation and Standards 494 Chapter 37 Grading A Summative Evaluation 514 PART VII REFERENCE SECTION 537 608 BIBLIOGRAPHY 538 552 APPENDIXES 553 584 GLOSSARY 585 595 INDEX 596 601 EPILOGUE 602 EPILOGUE A Standards for Sports Universities Departments illustrated 603 EPILOGUE B Standards for Sports Universities Courses illustrated 604 ABOUT THE AUTHOR 606 Readers Opinions Suggestions Form for Improvements in the Next Edition 607      Journal of Engineering Education ,2002      **The Architecture of Reform** Cary I. Sneider, Jacqueline Barber, Lincoln Bergman,1997      **Effects of Freedom to Control a Science Learning Sequence of Computer-based Instruction on Achievement and Attitudes of Junior High School Formal and Non-formal Reasoners** Nader Abdel-Aziz El-Sanhurry,1990      **Tests** Richard C. Sweetland, Daniel J. Keyser,1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services      *Thesaurus of ERIC Descriptors* ,1990 4th 7th eds contain a special

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web when the reactants have more potential energy than the products have and the extra energy is released to the surroundings it is called activation energy the height of the hill on a potential energy diagram is a measure of the endothermic the reaction above can be categorized as

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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring's unstretched length  $x = 0$  is  $U = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the  $pe$  of the reactants is 430kJ and the  $pe$  of the products is 190kJ what is the  $\Delta h$  2 if the  $pe$  of the reactants is 430kJ and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta h$  is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web answer see below explanation the rate of a chemical reaction  $a \rightarrow b \rightarrow c \rightarrow d$  is given in first approximation by  $v = k[A]^n$  where  $k = A e^{-E_a/RT}$  where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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web mar 17 2016 3 what is the activation energy of a reaction and how is this energy related to the activated complex of the reaction  $E_a$  is the minimum amount of energy for a reaction to occur it is the amount of energy required to create an activated complex 4 what happens when a catalyst is used in a reaction a catalyst changes the reaction

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy

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web key points potential energy is energy that has the potential to become another form of energy an object's potential energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object's mass and position in a gravitational field

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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web all questions are answered herein chemistry tutorial 9 02b potential energy diagrams how to draw and label pe diagrams for exothermic and endothermic reactions and what effect adding a catalyst or inhibitor has on the diagram

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web 1 potential energy diagrams are graphs energy is on the y axis and the x axis indicates geometry energy is a function of geometry 2 the geometry changes refer to changes in bonding changes in atom positions each point on the diagram has the same molecular formula same atoms and electrons the bonding can change however

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web may 1 2013 a potential energy diagram shows the change in potential energy of a system as reactants are converted into products figure below shows basic potential energy diagrams for an endothermic a and an exothermic b reaction recall that the enthalpy change  $\Delta h$  is positive for an endothermic reaction and negative for an

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web 1 draw a potential energy diagram for an endothermic reaction 2 label your drawing with the following letters a pe of

products d energy of activation b pe of reactants e heat of reaction c pe of the activated complex potential energy diagrams  
time ms 0 00 1 00 2 00 3 00 4 00 5 00 6 00 7 00 8 00 9 00 10 0 potential energy kcal 0 00 10 0

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web purdue chem 26100 organic chemistry i wenthold chapter 5 the study of chemical reactions

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web use the potential energy diagram to answer the questions below 1 is the overall reaction as shown exothermic or  
endothermic 2 what is the activation energy for the forward reaction 3 what is the

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nachteile das swing trading hat zeigen wir dir in diesem abschnitt welche vorteile haben swing

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sich als händler bewegt

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web der trader wartet darauf dass sich der dax dieser 200 tage linie annähert und von dort aus beginnt wieder anzusteigen das bietet die basis für einen swing trade long und den vorteil dass er den stop loss für diese position recht eng konkret ein bis zwei prozent unterhalb der 200 tage linie ansiedeln kann denn sollte dieser gleitende durchschnitt

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