Chapter 4 Practice Test - Answer Key

1) An element's properties can be predicted from its 48 a) number of isotopes b) number of neutrons c) atomic mass d) location in the periodic table The model of an atom is a ball of positive charge with negatively charged · c electrons embedded in it. a) Dalton b) Rutherford c) Thomson d) Bohr If a substance is _____, it can be pulled or drawn into a wire. 400 b) malleable c) magnetic d) reactive a) ductile 4) The spontaneous emission of radiation from an unstable atomic nucleus is called Do. a) corrosion b) radioactivity c) nuclear fusion d) reactivity et l 5) A(n) is a positively charged particle in an atom's nucleus. b) neutron c) plasma d) proton a) electron 6) Which of the following is NOT a characteristic of most metals? 28 a) brittle b) duetile c) good conductor d) malleable 7) An element's ______ tells the number of protons in its nucleus. lb. a) atomic mass b) atomic number c) chemical symbol d) period 8) Dmitri Mendeleev created the first **C** c) periodic table a) chemical reaction b) metal alloy d) semiconductor The particles that are lost, gained, or shared in chemical reactions are d b) protons c) neutrons d) electrons a) plasmas 10) A(n) ____c has some of the properties of both metals and nonmetals. a) alloy b) alkaline earth metal c) metalloid d) noble gas b (diatomic molecule) 11) A Tri-atomic Molecule consists of two nonmetal atoms bonded together. a) True b) false b (alkali metals) 12) The family of Alkaline Earth Metals is the most reactive group of metals in the periodic table. a) True b) False 13) The Lanthanides and the Actinides are shown at the bottom of the periodic table in order to keep the table a reasonable size. a) True b) False

Physical Science Test Answer Key Ch 6

Daniela Niemeyer

Physical Science Test Answer Key Ch 6:

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