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## Periodic Trends and Compounds Review Worksheet

**Keywords:** child sexual abuse; disclosure; self-blame

Use any material as your reference as you might deem necessary to answer the questions below in exception of asking someone else to give you the answer.

1. Which group tends to form 1+ ions? \_\_\_\_\_
2. Which group tends to form 2+ ions? \_\_\_\_\_
3. Which group tends to form 3+ ions? \_\_\_\_\_
4. Which group tends not to form ions or at all? \_\_\_\_\_

5. Locate the alkali metals, alkaline earth metals, halogens, noble gases, actinides, lanthanides, main group elements, semimetals (metalloids), transition metal s and inner transition metals on the periodic table below.

[illegible]

6. What repeating pattern is noticeable regarding the ionization energies of different elements? How does the periodic table account for this pattern?

7. Explain the patterns for acid ionizability by and discuss the reason the carboxylic acids

# Periodic Trends And Compounds Review

**Lei Shi**



## **Periodic Trends And Compounds Review:**

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*The Effects of Relativity in Atoms, Molecules, and the Solid State* Stephen Wilson, I.P. Grant, B.L. Gyorffy, 2012-12-06

Recent years have seen a growing interest in the effects of relativity in atoms molecules and solids On the one hand this can be seen as result of the growing awareness of the importance of relativity in describing the properties of heavy atoms and systems containing them This has been fueled by the inadequacy of physical models which either neglect relativity or which treat it as a small perturbation On the other hand it is dependent upon the technological developments which have resulted in computers powerful enough to make calculations on heavy atoms and on systems containing heavy atoms meaningful Vector processing and more recently parallel processing techniques are playing an increasingly vital role in rendering the algorithms which arise in relativistic studies tractable This has been exemplified in atomic structure theory where the dominant role of the central nuclear charge simplifies the problem enough to permit some prediction to be made with high precision especially for the highly ionized atoms of importance in plasma physics and in laser confinement studies Today's sophisticated physical models of the atom derived from quantum electrodynamics would be intractable without recourse to modern computational machinery Relativistic atomic structure calculations have a history dating from the early attempts of Swirles in the mid 1930's but continue to provide one of the primary test beds of modern theoretical physics

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