

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- A: Top left corner.
- B: Top right corner.
- C: Second row from top, second column from left.
- D: Second row from top, eighth column from left.
- E: Second row from top, tenth column from left.
- F: Third row from top, second column from left.
- G: Third row from top, sixth column from left.
- H: Third row from top, eighth column from left.
- I: Third row from top, tenth column from left.
- J: Fourth row from top, second column from left.
- K: Fourth row from top, tenth column from left.
- L: Fourth row from top, third column from left.
- M: Fourth row from top, sixth column from left.
- N: Fourth row from top, eighth column from left.
- O: Fourth row from top, tenth column from left.
- P: Fourth row from top, eleventh column from left.
- Q: Fifth row from top, second column from left.
- R: Fifth row from top, third column from left.
- S: Fifth row from top, sixth column from left.
- T: Fifth row from top, seventh column from left.
- U: Fifth row from top, eighth column from left.
- V: Bottom left corner of the second section.
- W: Bottom left corner of the third section.
- X: Bottom left corner of the fourth section.
- Y: Bottom left corner of the fifth section.

Arrows point from the letters R and Z to the letters V and X in the second section of the grid.

List elements in *alphabetical order* when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Periodic Tables Answer Key

Neil D. Jespersen, Pamela Kerrigan



Periodic Tables Answer Key:

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