

Percentage Yield Lab

Purpose: To determine the theoretical, actual and % yield of copper produced by the following reaction:



Materials and Method: Refer to Percentage Yield lab

Observations:

Mass of clean, empty beaker \rightarrow 78.5g
Mass of iron with beaker \rightarrow 79.255g
Filter paper \rightarrow 0.871g
Mass of copper inside filter paper \rightarrow 3.00g

Conclusions:

- 1) The mass of iron used is 0.755g because 79.255g (the weight of the iron filled beaker) - 78.5g (the weight of the empty beaker) = 0.755g
- 2) Fe + Cu
- 3) Two possible reasons which could explain the difference between your actual yield and the theoretical yield are the scale not being accurate and not being read. Have we waited long enough?

Percentage Yield Lab Grade 12

F Rizvi



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