Percent Composition and Empirical Formula

What is the percent composition of all the elements in Al₂(SO₄)₃?

2. A compound contains 36.5% Na, 25.4% S, and 38.1% O. Find its empirical formula.

- 3. Find the empirical formula of a compound consisting of 53.7% iron and 46.3% sulfur.
- 4. Determine the empirical formula for the following compounds:
- a. A compound containing 0.0130 mol carbon, 0.0390 mol hydrogen, and 0.0065 mol oxygen.

- b. A compound containing 72.7% magnesium and 27.8% nitrogen by mass.
- c. Glucose contains 40% carbon, 6.7% hydrogen, and 53.3% oxygen by mass.

Percent Composition And Empirical Formulas Answer Key

Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas **Percent Composition And Empirical Formulas Answer Key:**

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