Part B. Concept Review

Directions: Correctly complete each sentence by underlining the best of the three choices in parentheses.

1. (Physical changes, Density, Chemical changes) can be used to identify unknown compounds

and elements.

- 2. The states of matter are (acid, liquid, base), solid, gas, and plasma.
- 3. Density is a (size-dependent, size-independent, variable) property.
- Acids have a pH (between 0 and 7, exactly 7, between 7 and 14).
- Bases have a pH (between 0 and 7, exactly 7, between 7 and 14).
- You should (always, sometimes, never) taste or smell anything in the lab
- Crushing a leaf causes a (physical property, physical change, chemical change).
- A burning candle is an example of a (physical change, chemical property, chemical change).
- 9. A sign that a (physical change, chemical change, chemical property) has occurred is the
- release or absorption of energy.
- 10. When rocks are crushed, a (physical property, physical change, chemical change) has OCCULTED.

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