(b) P₁ > P₂ (c) P₁ < P₂ (d) P₂ = 0 Solution: (c) P₁ < P₂

Solution:

Numericals:

Convert 1mm of Hg into pascal. Take density of Hg=13.6 x 10³ kg m⁻³ and g=9.8 m/s²

Given: density of Hg=13.6 x 10^3 kg m⁻³, acceleration due to gravity =9.8 m/s², height of mercury column = 1mm or 0.001m

We know that pressure = h ρ g
= 0.001 x 13.6 x 10^3 x 9.8= 133.28 Pa

At a given place, a mercury barometer records a pressure of 0.70m of Hg. What would be the height of water column if mercury in barometer is replaced by water? Take density of mercury to be 13.6 x 10³ kg m⁻³

Given: density of Hg=13.6 x 10^3 kg m⁻³, acceleration due to gravity =9.8 m/s², height of mercury column = 0.70m We know that pressure = h ρ g $P = (0.7)(13.6 \times 10^3)(9.8)$ $= 93.3 \times 10^3 Pa$ Consider 'h' to the height of the water column P = h (density of water) g $93.3 \times 10^3 = h \times 1 \times 10^3 \times 9.8$ h = 93.39.8

At sea level, the atmospheric pressure is 76cm of Hg. If air pressure falls by 10mm of Hg
per 120m of ascent, what is the height of a hill where the barometer reads 70cm of Hg.
State the assumption made by you.

Solution: Given: atmospheric pressu

= 9.52 m

Given: atmospheric pressure = 76cm of Hg

Pressure falls by 10mm of Hg per 120m of ascent => 1cm of Hg per 120m of ascent

Consider 'h' to be the height of the hill, given=> pressure at the hill = 70cm of Hg

... total fall in pressure = atmospheric pressure – pressure at the hill

= 76 - 70 = 6cm of Hg

Given the rate of fall in pressure is 1cm Hg for every 120m advancement in height Hence for 6cm Hg, advancement in height is 6 x 120m = 720m Thus, the height of the hill is 720m The assumption made is that the atmospheric pressure falls linearly with ascent.

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