

### Quantum Theory

- a. Energy is quantized in  $\frac{h\nu}{2\pi}$  and  $\frac{h\nu}{2\pi}$  is the minimum energy
- b. Photo electron of energy is emitted as  $\frac{h\nu}{2\pi}$
- c. In quantized light energy is emitted as  $\frac{h\nu}{2\pi}$
- d. The energy of a photon of light depends on the  $\frac{h\nu}{2\pi}$  of light

$$E = \frac{h\nu}{2\pi}$$

\_\_\_\_\_ now the question is how to indicate this phenomenon

- a. Light energy of  $\frac{h\nu}{2\pi}$  is emitted as a certain amount of energy depending on the  $\frac{h\nu}{2\pi}$  of light
- b. Since  $\frac{h\nu}{2\pi}$  is the minimum energy photon of light
- c. It is also true that a photon of light of  $\frac{h\nu}{2\pi}$  is the minimum energy of light
- d. If light of  $\frac{h\nu}{2\pi}$  is emitted as a photon, then a photon of light of  $\frac{h\nu}{2\pi}$  is emitted
- e. If a photon of light of energy  $\frac{h\nu}{2\pi}$  is emitted, then a photon of light of energy  $\frac{h\nu}{2\pi}$  is emitted

What happens for light of energy of  $\frac{h\nu}{2\pi}$ ?

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What happens for light of energy of  $\frac{h\nu}{2\pi}$ ?

Light energy of  $\frac{h\nu}{2\pi}$  is emitted as a photon of light of energy  $\frac{h\nu}{2\pi}$

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**Rosina Ehmann**



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