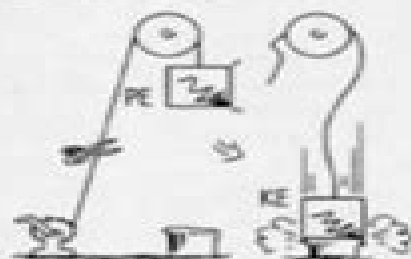
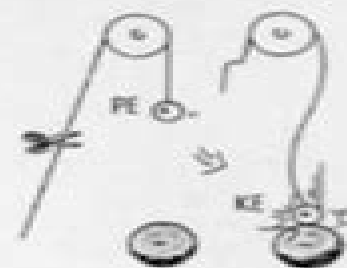


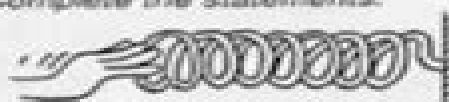
Chapter 21: Electrostatics
Electric Potential



Just as PE transforms to KE for a mass lifted against the gravitation field (left), the electric PE of an electric charge transforms to other forms of energy when it changes location in an electric field (right). In both cases, how does the KE acquired compare to the decrease in PE?



Complete the statements.



A force compresses the spring. The work done in compression is the product of the average force and the distance moved. $W = Fd$. This work increases the PE of the spring.

Similarly, a force pushes the charge (call it a test charge) closer to the charged sphere. The work done in moving the test charge is the product of the average _____ and the _____ moved. $W = \underline{\hspace{2cm}}$. This work _____ the PE of the test charge.



If the test charge is released, it will be repelled and fly past the starting point. Its gain in KE at this point is _____ to its decrease in PE.

At any point, a greater amount of test charge means a greater amount of PE. But not a greater amount of PE *per amount of charge*. The quantities PE (measured in joules) and PE/charge (measured in volts) are different concepts.

By definition: **Electric Potential = PE/charge**. 1 volt = 1 joule/1 coulomb. So 1 C of charge with a PE of 1 J has an electric potential of 1 V. And 2 C of charge with a PE of 2 J has an electric potential of _____ V.



If a conductor connected to the terminal of a battery has an electric potential of 12 V, then each coulomb of charge on the conductor has a PE of _____ J.



You do very little work in rubbing a balloon on your hair to charge it. The PE of several thousand billion electrons (about one-millionth coulomb [10^{-6} C]) transferred may be a thousandth of a joule [10^{-3} J]. Impressively, however, the electric potential of the balloon is about _____ V!

Why is contact with a balloon charged to thousands of volts not as dangerous as contact with household 110 V? _____

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Physics, Volume 2 David Halliday, Robert Resnick, Kenneth S. Krane, 2010-04-20 Written for the full year or three term Calculus based University Physics course for science and engineering majors the publication of the first edition of Physics in 1960 launched the modern era of Physics textbooks It was

a new paradigm at the time and continues to be the dominant model for all texts Physics is the most realistic option for schools looking to teach a more demanding course The entirety of Volume 2 of the 5th edition has been edited to clarify conceptual development in light of recent findings of physics education research End of chapter problem sets are thoroughly over hauled new problems are added outdated references are deleted and new short answer conceptual questions are added

Nuclear Science Abstracts ,1972 **The Electrical Review** ,1917 **Proceedings of COSPAR Symposium on Solar Particle Event of November 1969** James C. Ulwick,1972 A COSPAR sponsored Symposium on Solar Particle Events was held at Boston College at Chestnut Hill Massachusetts on 16 17 and 18 June 1971 The purpose of the symposium was to present measurements and observations made during the November 1969 Solar Proton Event This particular event was unique in that extensive solar magnetospheric and aeronomic measurements by rocket satellite and ground based techniques were made to provide a comprehensive picture of the phenomenology and morphology of this event In addition present knowledge of solar proton events and their effects on the polar ionosphere was reviewed and discussed This included theoretical studies as well as results from investigations of other solar proton events The symposium comprised invited as well as contributed papers A total of 49 papers presented at the symposium are included in these proceedings Applied Mechanics Reviews ,1975 Conductor Insulator Quantum Phase Transitions Vladimir Dobrosavljevic,Nandini Trivedi,James M. Valles (Jr),2012-06 When many particles come together how do they organize themselves And what destroys this organization Combining experiments and theory this book describes intriguing quantum phases metals superconductors and insulators and transitions between them It captures the excitement and the controversies on topics at the forefront of research **Chemistry and Industry** ,1973 *Western Electrician* ,1903 **The Pendulum** Gregory L. Baker,James A. Blackburn,2008-11-28 The pendulum a case study in physics is a unique book in several ways Firstly it is a comprehensive quantitative study of one physical system the pendulum from the viewpoint of elementary and more advanced classical physics modern chaotic dynamics and quantum mechanics In addition coupled pendulums and pendulum analogs of superconducting devices are also discussed Secondly this book treats the physics of the pendulum within a historical and cultural context showing for example that the pendulum has been intimately connected with studies of the earth s density the earth s motion and timekeeping While primarily a physics book the work provides significant added interest through the use of relevant cultural and historical vignettes This approach offers an alternative to the usual modern physics courses The text is amply illustrated and augmented by exercises at the end of each chapter **Power** ,1895 **Radio News** ,1939 Some issues 1943 July 1948 include separately paged and numbered section called Radio electronic engineering edition called Radionics edition in 1943 *Physics Briefs* ,1992 Radio & TV News ,1939-07 Some issues Aug 1943 Apr 1954 are called Radio electronic engineering ed called in 1943 Radionics ed which include a separately paged section Radio electronic engineering varies v 1 no 2 v 22 no 7 issued separately Aug 1954 May 1955 *Books and Pamphlets, Including Serials and*

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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web since the mole fraction of the o₂ is 0.050 it contributes 0.050 or if you will 5.0 of the total pressure thus total pressure is 0.21 0.050 4.2 atm problem 9 a sample of 1.43 g of helium and an unweighed quantity of o₂ are mixed in a flask at room temperature