

Physical Vs. Chemical Properties

A physical property is observed with the senses and can be determined without changing the object. Examples: color, shape, mass, length and more are all characteristics of physical properties.

A chemical property indicates how a substance reacts with something else. The simplest example is flammability, a chemical property of substances. For example, the ability of wood to burn is a chemical property. The wood is converted into oxygen, and the original wood is destroyed. In some cases, even the number of different substances with a different formula: $2H_2 + O_2 \rightarrow 2H_2O$.

Identify the following properties as either chemical or physical by putting a check in the appropriate column.

	Physical Property	Chemical Property
1. color change	<input type="checkbox"/>	<input type="checkbox"/>
2. odor change	<input type="checkbox"/>	<input type="checkbox"/>
3. temperature change	<input type="checkbox"/>	<input type="checkbox"/>
4. solubility	<input type="checkbox"/>	<input type="checkbox"/>
5. mass or volume before and after	<input type="checkbox"/>	<input type="checkbox"/>
6. appearance or composition	<input type="checkbox"/>	<input type="checkbox"/>
7. pressure	<input type="checkbox"/>	<input type="checkbox"/>
8. melting point	<input type="checkbox"/>	<input type="checkbox"/>
9. boiling point or change in volume or gas	<input type="checkbox"/>	<input type="checkbox"/>
10. reacts with a liquid to form another	<input type="checkbox"/>	<input type="checkbox"/>
11. combusts	<input type="checkbox"/>	<input type="checkbox"/>
12. freezing point	<input type="checkbox"/>	<input type="checkbox"/>
13. solubility in water	<input type="checkbox"/>	<input type="checkbox"/>
14. density	<input type="checkbox"/>	<input type="checkbox"/>
15. conductivity	<input type="checkbox"/>	<input type="checkbox"/>

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