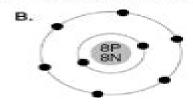
# IPS Unit 9 – Chemical Bonds Types of Bonds Worksheet

Directions: Study the diagram below. Write your answers to the questions in the spaces provided.

((2P))



- 1. If atom A loses electrons to atom B,
  - a. how many electrons will atom A lose?
  - b. how many electrons will atom B gain? \_\_\_\_\_
  - e. what will be the charge of atom A?
  - d. what will be the charge of atom B?
  - e. what will be the total charge of the compound formed?
  - f. what type of bond will form?
- 2. Explain why an element's charge is related to the group on the periodic table to which it belongs.

Directions: Complete the table comparing ionic compounds and covalent compounds.

Characteristic	Ionic compounds	Covalent compounds
How the compound is formed		
Smallest particle		
Usual state at room temperature		

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**WJ Hussar** 

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