

## **Physical Science- Semester 1 Study Guide**

### **Chapter 1- Nature of Science**

|                   |            |                      |
|-------------------|------------|----------------------|
| Scientific method | hypothesis | experiment           |
| variable          | constant   | dependent variable   |
| control           | model      | independent variable |
| theory            | technology | scientific law       |
| volume            | mass       | density              |
| graph             |            |                      |

### **Chapter 18- Classification of Matter**

|                             |                |                       |
|-----------------------------|----------------|-----------------------|
| substance                   | element        | heterogeneous mixture |
| compound                    | solution       | homogeneous mixture   |
| colloid                     | Tyndell effect | suspension            |
| physical property           | distillation   | chemical property     |
| physical change             |                | chemical change       |
| law of conservation of mass |                |                       |

### **Chapter 19- Properties of Atoms & the Periodic Table**

|                |                  |                      |
|----------------|------------------|----------------------|
| atom           | nucleus          | proton               |
| neutron        | electron         | quark                |
| electron cloud | valance electron | atomic number        |
| mass number    | isotope          | average atomic mass  |
| periodic table | group            | electron dot diagram |
| period         |                  |                      |

### **Chapter 22- Chemical Bonds**

|                  |                |                   |
|------------------|----------------|-------------------|
| chemical formula | ion            | noble gases       |
| chemical bond    | ionic bond     | covalent bond     |
| molecule         | polar molecule | nonpolar molecule |
| binary compound  | hydrate        | oxidation number  |
| polyatomic ion   |                |                   |

### **Chapter 23- Chemical Reactions**

|                              |             |                              |
|------------------------------|-------------|------------------------------|
| chemical reaction            | reactant    | product                      |
| chemical equation            | coefficient | balanced chemical equation   |
| combustion reaction          | synthesis   | decomposition reaction       |
| single-displacement reaction |             | double-displacement reaction |
| activation energy            | catalyst    | endothermic reaction         |
| exothermic reaction          | inhibitor   | rate of reaction             |

# Physical Science Semester 1 Study Guide

**Michael Seilmaier**



## **Physical Science Semester 1 Study Guide:**

## Whispering the Secrets of Language: An Psychological Quest through **Physical Science Semester 1 Study Guide**

In a digitally-driven earth wherever monitors reign supreme and quick conversation drowns out the subtleties of language, the profound secrets and emotional subtleties hidden within phrases usually get unheard. However, located within the pages of **Physical Science Semester 1 Study Guide** a fascinating literary prize sporting with natural feelings, lies an extraordinary quest waiting to be undertaken. Penned by an experienced wordsmith, this charming opus invites readers on an introspective journey, delicately unraveling the veiled truths and profound impact resonating within the very fabric of every word. Within the mental depths of the moving evaluation, we will embark upon a honest exploration of the book is primary styles, dissect its fascinating writing style, and fail to the effective resonance it evokes heavy within the recesses of readers hearts.

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