

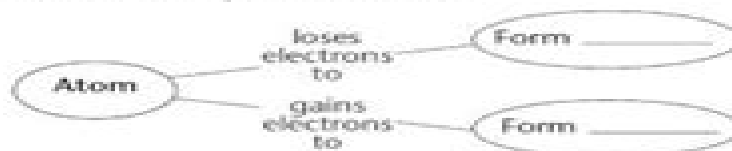
Chapter 6 Chemical Bonds

Section 6.1 Ionic Bonding

(pages 158–164)

*This section describes the formation of ionic bonds and the properties of ionic compounds.***Reading Strategy** (page 158)

Sequencing As you read, complete the concept map to show what happens to atoms during ionic bonding. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Stable Electron Configurations** (page 158)

1. Circle the letter that describes the type of electron configuration that makes an atom stable and not likely to react.
 - a. The lowest energy level of an atom is filled.
 - b. The highest energy level of an atom is filled.
 - c. All the energy levels of an atom are filled.
2. Define an electron dot diagram. _____

Ionic Bonds (pages 159–161)

3. Some elements achieve stable electron configurations through the transfer of _____ between atoms.
4. Circle the letter that states the result of a sodium atom transferring an electron to a chlorine atom.
 - a. Each atom ends up with a more stable electron arrangement.
 - b. The sodium atom becomes more stable, but the chlorine atom becomes less stable.
 - c. The chlorine atom becomes more stable, but the sodium atom becomes less stable.

Physical Science Reading And Study Workbook Chapter 6 2

J Rink



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