

NUCLEAR CHEMISTRY WORKSHEET

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Most nuclear reactions involve changes in the number of protons and/or neutrons in the nucleus of an atom. These changes are called *transmutations* because an atom of one element is changed into an atom of a different element. During these changes, nuclear particles may be absorbed and/or emitted along with a release of energy.

There are four types of nuclear reactions:

- 1) *natural radioactive decay*: transmutations occur by natural radioactivity (the ability of a nucleus to emit a nuclear particle and energy without external stimulation);
- 2) *artificial transmutation*: transmutations occur by nuclear disintegration caused by external stimulation as a scientist bombards a nucleus with a particle (the addition of another nuclear particle makes the nucleus unstable);
- 3) *fission*: certain nuclei having a large mass are bombarded with special particles that cause the nuclei to split into two nuclei each having a smaller mass;
- 4) *fusion*: nuclei of light elements are combined to form heavier nuclei.

Nuclear reactions are written using symbols in the A_ZX notation. The symbols for the major nuclear particles involved in nuclear reactions are given below.

Alpha particle	${}^4_2\text{He}$ (an alpha particle is a helium nucleus)
Proton	${}^1_1\text{H}$ (the most common hydrogen nucleus is a proton)
Neutron	${}^1_0\text{n}$
Electron	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ (also called a beta particle)
Positron	${}^0_1\text{e}$ or ${}^0_1\beta$
Gamma ray	${}^0_0\gamma$

The table below contains information about nuclear reactions involving the emission of radiation.

Type of Reaction	NUCLEAR CHANGES		
	Radiation (Particle)	Effect on the Atomic Number	Effect on the Atomic Mass
Alpha emission (α)	${}^4_2\text{He}$	decrease by 2	decrease by 4
Beta emission (β)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	increase by 1	no change
Positron emission (β^+)	${}^0_1\text{e}$ or ${}^0_1\beta$	decrease by 1	no change
Gamma emission	${}^0_0\gamma$	no change	no change
Electron capture (EC)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	decrease by 1	no change

Physical Scienceworkbook Chapter Nuclear Chemistry

Wordwise Answers

Xiang Xie



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Table of Contents Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers

1. Understanding the eBook Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - The Rise of Digital Reading Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Personalized Recommendations
 - Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers User Reviews and Ratings
 - Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers and Bestseller Lists

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 - Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers eBook Subscription Services
 - Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers Compatibility with Devices
 - Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Highlighting and Note-Taking Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Interactive Elements Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
8. Staying Engaged with Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
9. Balancing eBooks and Physical Books Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Setting Reading Goals Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Fact-Checking eBook Content of Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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