

Physical Science- Semester 1 Study Guide

Chapter 1- Nature of Science

Scientific method	hypothesis	experiment
variable	constant	dependent variable
control	model	independent variable
theory	technology	scientific law
volume	mass	density
graph		

Chapter 18- Classification of Matter

substance	element	heterogeneous mixture
compound	solution	homogeneous mixture
colloid	Tyndell effect	suspension
physical property	distillation	chemical property
physical change		chemical change
law of conservation of mass		

Chapter 19- Properties of Atoms & the Periodic Table

atom	nucleus	proton
neutron	electron	quark
electron cloud	valance electron	atomic number
mass number	isotope	average atomic mass
periodic table	group	electron dot diagram
period		

Chapter 22- Chemical Bonds

chemical formula	ion	noble gases
chemical bond	ionic bond	covalent bond
molecule	polar molecule	nonpolar molecule
binary compound	hydrate	oxidation number
polyatomic ion		

Chapter 23- Chemical Reactions

chemical reaction	reactant	product
chemical equation	coefficient	balanced chemical equation
combustion reaction	synthesis	decomposition reaction
single-displacement reaction		double-displacement reaction
activation energy	catalyst	endothermic reaction
exothermic reaction	inhibitor	rate of reaction

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