#### Trends on the Periodic Table

#### Part 1: Identifying chemical properties/characteristics of the elements

For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.)



# Atomic Rodii Size:

1. Explain atomic radius: A measure of the size (volume) of an atom. Determined by measuring half of the distance between two identical atoms bonded together.

2. Using the information in figure 5-13 (pg 141) in your book, write the atomic radii size for alkali metals & period #2 on the periodic table to your right.



- 3. What happens to atomic radii as you move down a column and across a row? On the arrows on the diagram, indicate the trend. (Options: gets bigger OR gets smaller)
- 4. Of all the elements on the periodic table, identify & circle the largest & smallest in size.
- Rank the following elements from largest to smallest in atomic radii: a) Be, Ba, Mg = Ba, Mg, Be
- 6. How can you explain the trend:

a. Why do they get smaller as you go across a period? Additional protons in the nucleus "pull" on

in creases the attraction of the nucleus and therefore cloud close to the nucleus resulting

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In a digitally-driven world wherever monitors reign great and immediate communication drowns out the subtleties of language, the profound techniques and mental subtleties hidden within words often move unheard. However, situated within the pages of **Periodic Trends Pogil Activities For High School Chemistry Answers** a fascinating literary treasure sporting with raw emotions, lies a fantastic journey waiting to be undertaken. Published by an experienced wordsmith, this charming opus invites visitors on an introspective journey, delicately unraveling the veiled truths and profound influence resonating within the very fabric of each and every word. Within the emotional depths with this emotional review, we shall embark upon a honest exploration of the book is core styles, dissect their charming publishing design, and fail to the strong resonance it evokes deep within the recesses of readers hearts.

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