

# GRAPHING TRENDS OF THE PERIODIC TABLE

## CHEMISTRY

NAME: \_\_\_\_\_

DATE: \_\_\_\_\_

PERIOD: \_\_\_\_\_

### Graphing Periodic Table Trends

**Directions:** Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	157	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	175	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	145
Nitrogen - 7	55	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

### Atomic Radius



# Periodic Trends Graphing And Questions Key

**Princeton Review**



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## **Table of Contents Periodic Trends Graphing And Questions Key**

1. Understanding the eBook Periodic Trends Graphing And Questions Key
  - The Rise of Digital Reading Periodic Trends Graphing And Questions Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Graphing And Questions Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Trends Graphing And Questions Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Graphing And Questions Key
  - Personalized Recommendations
  - Periodic Trends Graphing And Questions Key User Reviews and Ratings
  - Periodic Trends Graphing And Questions Key and Bestseller Lists
5. Accessing Periodic Trends Graphing And Questions Key Free and Paid eBooks
  - Periodic Trends Graphing And Questions Key Public Domain eBooks
  - Periodic Trends Graphing And Questions Key eBook Subscription Services
  - Periodic Trends Graphing And Questions Key Budget-Friendly Options
6. Navigating Periodic Trends Graphing And Questions Key eBook Formats

- ePub, PDF, MOBI, and More
- Periodic Trends Graphing And Questions Key Compatibility with Devices
- Periodic Trends Graphing And Questions Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Trends Graphing And Questions Key
  - Highlighting and Note-Taking Periodic Trends Graphing And Questions Key
  - Interactive Elements Periodic Trends Graphing And Questions Key
- 8. Staying Engaged with Periodic Trends Graphing And Questions Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Trends Graphing And Questions Key
- 9. Balancing eBooks and Physical Books Periodic Trends Graphing And Questions Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Trends Graphing And Questions Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Graphing And Questions Key
  - Setting Reading Goals Periodic Trends Graphing And Questions Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Graphing And Questions Key
  - Fact-Checking eBook Content of Periodic Trends Graphing And Questions Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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