

## 2.6 - Periodic Trends worksheet

Name \_\_\_\_\_

### I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

### II. Answer the questions using the symbols on the periodic table

\_\_\_\_\_ 1. What elements are in period 5?

What element(s) is a

\_\_\_\_\_ 2. halogens

\_\_\_\_\_ 3. lanthanides

\_\_\_\_\_ 4. alkaline earth metals

\_\_\_\_\_ 5. noble gases

\_\_\_\_\_ 6. transition metals

\_\_\_\_\_ 7. alkali metals

\_\_\_\_\_ 8. actinides

What elements are in the sublevel

\_\_\_\_\_ 9. s

\_\_\_\_\_ 10. p

\_\_\_\_\_ 11. d

\_\_\_\_\_ 12. f

What elements are

\_\_\_\_\_ 13. metals,

\_\_\_\_\_ 14. nonmetals

\_\_\_\_\_ 15. metalloids,

\_\_\_\_\_ 16. Which element is the most reactive nonmetal?

\_\_\_\_\_ 17. Which element is the most reactive metal?

\_\_\_\_\_ 18. Which element below is the least reactive?

\_\_\_\_\_ 19. Which element has the largest radius in period 4?

\_\_\_\_\_ 20. Which element has the largest radius in the Alkali metals family?

\_\_\_\_\_ 21. Which element has the highest ionization energy in period four?

\_\_\_\_\_ 22. Which elements has the second largest atomic number?

\_\_\_\_\_ 23. Which element has the largest atomic mass?

\_\_\_\_\_ 24. Which element is located at period 4, group 15?

\_\_\_\_\_ 25. Which element is located at period 5, group 2?

\_\_\_\_\_ 27. Which elements will form +1 ions?

\_\_\_\_\_ 28. Which group of metals will have the same properties

\_\_\_\_\_ 29. Which group of nonmetals will have the same properties

\_\_\_\_\_ 30. On the back, define all chapter 5, section 3 vocabulary terms.

b																	r		
																	h		o
	a																		u
i								t											
				e														z	

		w												
					y									



# Periodic Table Chemistry Answers

**Yan Bai**



## **Periodic Table Chemistry Answers:**

Chemistry Questions - Answers for Intermediate Mr. Ajeet Singh, 2018-05-13 A question answer formatted book designed for intermediate level chemistry students covering theory based and numerical questions to aid learning and exam preparation

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