



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Problems Answers

Michael Brown

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Percent Yield Problems Answers:

Whispering the Techniques of Language: An Emotional Journey through **Percent Yield Problems Answers**

In a digitally-driven world wherever monitors reign supreme and immediate transmission drowns out the subtleties of language, the profound strategies and mental subtleties hidden within phrases often move unheard. However, located within the pages of **Percent Yield Problems Answers** a captivating literary treasure blinking with raw emotions, lies an extraordinary journey waiting to be undertaken. Written by an experienced wordsmith, this charming opus encourages readers on an introspective trip, gently unraveling the veiled truths and profound impact resonating within the fabric of every word. Within the psychological depths of this poignant evaluation, we will embark upon a heartfelt exploration of the book is key styles, dissect its fascinating publishing type, and fail to the strong resonance it evokes strong within the recesses of readers hearts.

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