

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Which statement concerning Arrhenius acid-base theory is **not** correct? 1) _____
A) Acid-base reactions must take place in aqueous solution.
B) An Arrhenius acid produces hydrogen ions in water solution.
C) An Arrhenius base produces hydroxide ions in water solution.
D) A neutralization reaction produces water plus a salt.
E) none of the above
- 2) The H_3O^+ ion is called the _____ ion. 2) _____
A) protium
B) hydronium
C) hydrogen
D) hydroxide
E) water
- 3) When acids and bases react the product other than water is a 3) _____
A) hydroxide ion.
B) hydrogen ion.
C) hydronium ion.
D) salt.
E) metal.
- 4) Which of the following compounds is a salt? 4) _____
A) NaOH B) KNO_3 C) $\text{C}_6\text{H}_{12}\text{O}_6$ D) H_2SO_4 E) HBr
- 5) Which of the following compounds is a salt? 5) _____
A) $\text{CH}_3\text{CO}_2\text{H}$
B) NH_4NO_3
C) C_6H_6
D) NH_3
E) $\text{Al}(\text{OH})_3$
- 6) Which compound is manufactured in larger quantities in the U.S. than any other industrial chemical? 6) _____
A) HNO_3 B) H_2SO_4 C) HCl D) H_3PO_4 E) NaOH

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