Unit 2: The Periodic Table Review Sheet

UZLI: PT History

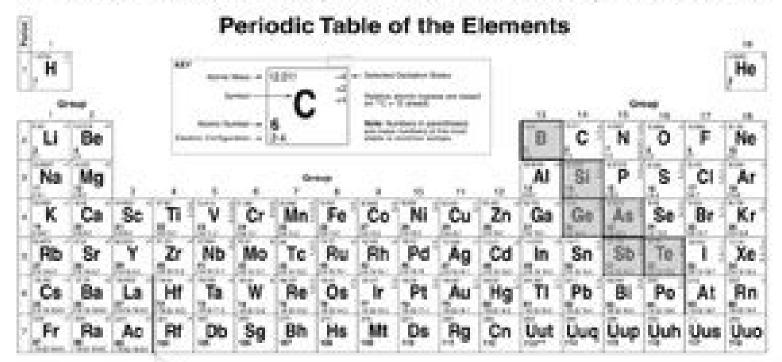
- On the present day Periodic Table, elements are arranged in order of increasing atomic number
- Groups: vertical columns on the periodic table
 - Elements in the same group have similar chemical properties because they have the same number of valence electrons
 - Valence electron: outermost electrons (lost # in the electron configuration)
- Periods: horizontal rows on the periodic table.

U21.2: Ports of the Periodic Table

Elements are classified as either metals, nanmetals, or metalloids.

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Classification can be determined by location on the periodic table or by different properties.



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Periodic Table Review Pt 2

The Princeton Review

Periodic Table Review Pt 2:

Periodic Table Review Pt 2 Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Periodic Table Review Pt 2**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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