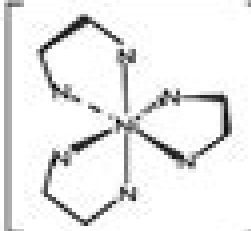
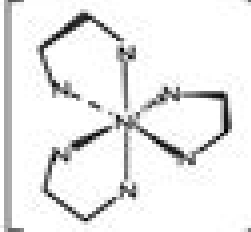
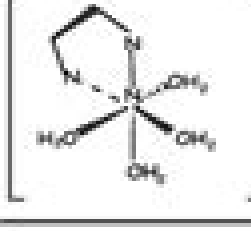
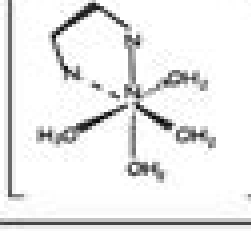


Table 1: Synthesis of an Unknown Complex Nickel Salt

Unknown	Formula	Both ethylenediamine and waters form covalent bonds with the Ni^{2+} by donating a lone pair. Ethylenediamine can forms two covalent bonds, while water will only form one. $\text{NH}_2\text{-CH}_2\text{-CH}_2\text{-NH}_2 = (\text{en})$	Molar Mass
Hypothesis A	$[\text{Ni}(\text{en})_3]\text{SO}_4 \cdot 5\text{H}_2\text{O}$	 SO_4 + 5 H_2O	424.99 g/mol
Hypothesis B	$[\text{Ni}(\text{en})_3]\text{SO}_4$	 SO_4	334.92 g/mol
Hypothesis C	$[\text{Ni}(\text{H}_2\text{O})_4(\text{en})]\text{SO}_4$	 SO_4	286.78 g/mol
Hypothesis D	$[\text{Ni}(\text{H}_2\text{O})_4(\text{en})]\text{SO}_4 \cdot 5\text{H}_2\text{O}$	 SO_4 + 5 H_2O	376.86 g/mol

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