

PERIODIC PROPERTIES LAB

Objective: To determine the relationship between the reactivity of elements and their location on the Periodic Table.

Procedure:**Part I: Halogens**

1. Put 5mL of NaCl into a test tube.
2. Add HNO₃ to the test tube one drop. After each drop test to see if the solution turns the blue litmus paper pink. (Test this by dipping the stirring rod into the test tube and then touching it to the litmus paper).
3. After the blue litmus paper turns pink, add 5 more drops of HNO₃.
4. Add 5mL of AgNO₃ to the test tube. Record observations.
5. Repeat steps 1-4 for NaBr, NaI and the unknown halogen in CLEAN test tubes.

Part II: Alkaline Earth Metal Salts

1. Put 1mL of each into their own test tubes: Mg(NO₃)₂, Ca(NO₃)₂, Sr(NO₃)₂, and Ba(NO₃)₂.
2. Add 1mL of 1M H₂SO₄ to each test tube. Record observations.
3. Rinse test tubes and refill with the original solutions.
4. Add 5mL of 1M Na₂CO₃ to each. Record observations.
5. Rinse, refill, and add 5mL of 0.25M (NH₄)₂C₂O₄ to each. Record observations.
6. Repeat with unknown alkaline earth metal salt.

Part I: Data:

Halide	Observations
NaCl	
NaBr	
NaI	
Unknown	

Periodic Properties Lab Answer Key

Effiong Eyo

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the name bar.

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Table of Contents Periodic Properties Lab Answer Key

1. Understanding the eBook Periodic Properties Lab Answer Key
 - The Rise of Digital Reading Periodic Properties Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Properties Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Properties Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Properties Lab Answer Key
 - Personalized Recommendations
 - Periodic Properties Lab Answer Key User Reviews and Ratings
 - Periodic Properties Lab Answer Key and Bestseller Lists
5. Accessing Periodic Properties Lab Answer Key Free and Paid eBooks

- Periodic Properties Lab Answer Key Public Domain eBooks
- Periodic Properties Lab Answer Key eBook Subscription Services
- Periodic Properties Lab Answer Key Budget-Friendly Options
- 6. Navigating Periodic Properties Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Properties Lab Answer Key Compatibility with Devices
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- 7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Periodic Properties Lab Answer Key
 - Interactive Elements Periodic Properties Lab Answer Key
- 8. Staying Engaged with Periodic Properties Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Properties Lab Answer Key
- 9. Balancing eBooks and Physical Books Periodic Properties Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Properties Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Properties Lab Answer Key
 - Setting Reading Goals Periodic Properties Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Properties Lab Answer Key
 - Fact-Checking eBook Content of Periodic Properties Lab Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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