Question		Answer	Marks	Guidance
	(a)	F <sub>2</sub> forces mark F <sub>2</sub> has van der Waals' (forces) OR F <sub>2</sub> has induced dipole attractions OR interactions OR F <sub>3</sub> has temporary OR instantaneous dipole(-dipole) attraction OR interactions   //	4	Use annotations is ticks crosses ECF * etc for this part  ALLOW vdWs for van der Waals' IGNORE F <sub>2</sub> has covalent bond for this mark IGNORE F <sub>2</sub> has 'intermolecular forces'  Quality of written communication: 'dipole(s)' spelled correctly and used in context for the second marking point IGNORE HC! has 'intermolecular forces' IGNORE van der Waals' forces in HC!
		Comparison of strength of forces between molecules mark intermolecular force in HCI is stronger than that in F <sub>2</sub> OR permanent dipoles are stronger (than induced dipoles) <		DO NOT ALLOW hydrogen bonding  Look for strength of force comparison anywhere in the answer  ALLOW ECF for hydrogen bonding in HCI being stronger than the stated intermolecular forces in F <sub>2</sub> BUT DO NOT ALLOW this mark if HCI or F <sub>2</sub> has covalent bonds broken OR if HCI has ionic bonds broken (the
		Boiling point mark more energy is required to break stronger (intermolecular) forces <		question asks for forces between molecules) IGNORE HC/ has stronger van der Waals' (forces) than F <sub>2</sub> (as they both have the same number of electrons)  DO NOT ALLOW fourth mark if covalent bonds are broken in HC/ or F <sub>2</sub> OR if ionic bonds are broken in HC/ IGNORE 'heat' but ALLOW 'heat energy'

# Ocr F321 Jan 2013 Paper

**Peter Cann, Peter Hughes** 

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