

Question number	Answer	Marks	Guidance
6 (c)	$M(\text{NaNO}_3) = 85.0 \text{ g mol}^{-1}$ $n(\text{NaNO}_3) = 0.04 \text{ mol}$ $n(\text{O}_2) = 0.02 \text{ mol}$ $\text{Volume of O}_2 = 0.0200 \times 24\,000 = 480 \text{ cm}^3$	B1 B1 B1 B1	
7 (a)	$0.0250 \times 23.0 = 0.575 \text{ g}$	B1	
7 (b)	$2\text{Na(s)} + 2\text{H}_2\text{O(l)} \rightarrow 2\text{NaOH(aq)} + \text{H}_2\text{(g)}$	B1	
7 (c)	$n(\text{H}_2) = 0.0125 \text{ mol}$ $\text{Volume H}_2 = 0.0125 \times 24\,000 = 300 \text{ cm}^3$	B1 B1	
7 (d) (i)	$n(\text{NaOH}) = n(\text{Na}) = 0.0250 = c \times 50/1000$ $c = 0.500 \text{ mol dm}^{-3}$	B1	
7 (d) (ii)	$c = 0.500 \times 40.0 = 20.0 \text{ g dm}^{-3}$	B1	
8 (a)	58.5 g	B1	
8 (b)	$n(\text{Cl}_2) = (2.5 \times 10^3)/24 = 1.04 \times 10^2 \text{ mol}$ $n(\text{NaOH}) = 2 \times 1.04 \times 10^2 = 2.08 \times 10^2 \text{ mol}$ $2.08 \times 10^2 = 4.00 \times V \text{ (in dm}^3\text{)}$ $\therefore V = 5.2 \times 10^7 \text{ dm}^3$	B1 B1 B1	
9 (a) (i)	$\text{C} : \text{H} = 54.55/12.0 : 9.09/1.0 : 36.36/16.0 = 4.55 : 9.09 : 2.27$ $\text{Empirical formula} = \text{C}_2\text{H}_4\text{O}$	B1 B1	
9 (a) (ii)	$pV = nRT$ $n = \frac{(103 \times 10^3) \times (72.0 \times 10^{-6})}{8.314 \times 373} =$ $0.002\,39 \text{ mol}$ $M = \frac{0.2103}{0.00239} = 88.0$	B1 B1 B1 B1	
9 (a) (iii)	$\text{Molecular formula} = \text{C}_2\text{H}_4\text{O} \times 88/44 = \text{C}_4\text{H}_8\text{O}_2$	B1	
10 (a) (i)	Al^{3+} SO_4^{2-}	B1 B1	

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1. Understanding the eBook Ocr Chemistry As Answers
 - The Rise of Digital Reading Ocr Chemistry As Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry As Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Chemistry As Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry As Answers
 - Personalized Recommendations

- Ocr Chemistry As Answers User Reviews and Ratings
- Ocr Chemistry As Answers and Bestseller Lists
- 5. Accessing Ocr Chemistry As Answers Free and Paid eBooks
 - Ocr Chemistry As Answers Public Domain eBooks
 - Ocr Chemistry As Answers eBook Subscription Services
 - Ocr Chemistry As Answers Budget-Friendly Options
- 6. Navigating Ocr Chemistry As Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr Chemistry As Answers Compatibility with Devices
 - Ocr Chemistry As Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry As Answers
 - Highlighting and Note-Taking Ocr Chemistry As Answers
 - Interactive Elements Ocr Chemistry As Answers
- 8. Staying Engaged with Ocr Chemistry As Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Chemistry As Answers
- 9. Balancing eBooks and Physical Books Ocr Chemistry As Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Chemistry As Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr Chemistry As Answers
 - Setting Reading Goals Ocr Chemistry As Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr Chemistry As Answers
 - Fact-Checking eBook Content of Ocr Chemistry As Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web feb 17 2017 la cuisine japonaise étant très imprégnée des fondements de la religion bouddhiste manger des produits de saison est non seulement un gage de plats plus savoureux mais également de respect envers la nature en s adaptant à son rythme pour illustration le calendrier traditionnel japonais liste 24 saisons