Graphing Periodic Trends Worksheet

 Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

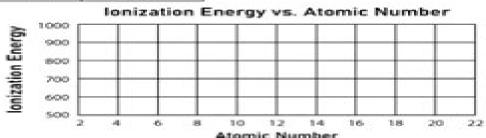
			Period 2
Element	Atomic Number	IE (kJ/mol)	
Li	3	519	
Вe	4	900	
В	5	799	
C	6	1088	

Element	Atomic Number	IE (kJ/mol)
Li	3	519
Be	4	900
В	5	799
C	6	1088

2500								
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L								

- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
- Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A				
Element	Atomic Number	IE (kJ/mol)		
Be	4	900		
Mg	12	736		
Ca	20	590		



Reading Periodic Trend Graphing Questions

Linda Froschauer, Mary L. Bigelow

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