

10.4 Electrophilic Reactions

General Introduction An electrophile is an electron-seeking species, reacting with a lone pair or a double bond.

Electrophilic Substitution

Electrophilic Substitution reactions are reactions in the aromatic ring with and without

- is a reaction in which a group on the benzene ring is replaced by an electrophile.
- The electrophile can be a cation or a molecule, such as H^+ or H_2O , which is then replaced by a leaving group.

Electrophilic Substitution reactions are classified by the mechanism of the reaction.

- The first mechanism involves the electrophile attacking the benzene ring, forming a carbocation intermediate.
- The second mechanism involves the electrophile attacking the benzene ring, forming a carbocation intermediate.

When reacting benzene with H^+ , the following reaction occurs:

Electrophilic Substitution

- In the following mechanism, the electrophile H^+ attacks the benzene ring, forming a carbocation intermediate. The leaving group is H_2O . _____
- Write down the mechanism for the reaction of benzene with H^+ . _____
- Explain why the reaction of benzene with H^+ is an electrophilic substitution reaction. _____
- Write down the mechanism for the reaction of benzene with H^+ . _____
- Write down the mechanism for the reaction of benzene with H^+ . _____
- Write down the mechanism for the reaction of benzene with H^+ . _____

Transition

- Write down the mechanism for the reaction of benzene with H^+ . _____

Reading And Study Workbook 19 4 Neutralization Reactions

H Kauffman



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