

Trends on the Periodic Table

Part 1: Identifying chemical properties/characteristics of the elements

For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.)

Lithium
Being one of the smaller elements with only 3 electrons, lithium is happy to give one of its electrons away and never seems to ask for any in return.



Francium
Francium is the largest of all of the elements and has so many electrons that it has trouble keeping track of them. Francium will eagerly give up electrons to other elements and never takes in extras. But watch out, it has an extremely reactive and explosive personality!

Fluorine
One of the smallest elements, Fluorine is often described as snappy with a "magnetic" personality. As part of the halogen family (outermost electron shell), Fluorine is the best at convincing other elements to give up their electrons to him. However, he holds on tight to his own electrons and will not let them go without a fight.



Iodine
As part of the halogen family, Iodine is also a skilled electron thief, however, not nearly as good as its smaller brothers and sisters. Iodine will also put up a pretty good fight to keep its own electrons.



Neon
Part of the Noble family, Neon does not readily interact with other elements. With a full set of valence electrons, Neon feels that he has everything he needs to make him happy without sharing with others.

Part 2: Use your textbook to complete the following:

Atomic Radii Size:

1. Explain **atomic radius**: A measure of the size (volume) of an atom. Determined by measuring half of the distance between two identical atoms bonded together.
2. Using the information in figure 5-13 (pg 141) in your book, **write** the atomic radii size for **alkali metals & period #2** on the periodic table to your right.



3. What happens to atomic radii as you move down a column and across a row? On the arrows on the diagram, indicate the trend. (Options: gets bigger OR gets smaller)
4. Of all the elements on the periodic table, **identify & circle the largest & smallest** in size.

5. Rank the following elements from largest to smallest in atomic radii:

a) Be, Ba, Mg = **Ba, Mg, Be**

b) Li, F, B = **Li, B, F**

6. How can you explain the trend?

a. Why do they get smaller as you go across a period? Additional protons in the nucleus "pull" on electrons, bring the electron cloud closer.

Effective nuclear charge, Z_{eff} , increases the attraction of the nucleus and therefore pulls the electron cloud closer to the nucleus resulting in smaller radii.

*not a good enough answer
4AP Answer*

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is using the Review Book as instructional material SHOULD NOT buy the Home Edition Also available in paperback print

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order **Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Kaplan SAT Subject Test Chemistry 2015-2016* Kaplan Test Prep, 2015-03-03 Essential strategies practice and review to ace the SAT Subject Test Chemistry Getting into a top college has never been more difficult Students need to distinguish themselves from the crowd and scoring well on a SAT Subject Test gives students a competitive edge Kaplan's SAT Subject Test Chemistry is the most up to date guide on the market with complete coverage of both the content review and strategies students need for success on test day Kaplan's SAT Subject Test Chemistry features A full length diagnostic test Full length practice tests Focused chapter summaries highlights and quizzes Detailed answer explanations Proven score raising strategies End of chapter quizzes Kaplan is serious about raising students scores we guarantee students will get a higher score *Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter *Holt*

Chemistry Ralph Thomas Myers, 2004 *Chemistry* John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14
 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind. A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings. It more accurately reflects the curriculum of most Canadian institutions. Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant.

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000. The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry. Designed to help readers master the principles of general chemistry. As a prep book it promotes active involvement with the material. There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills. Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems.

Chemical Principles Peter Atkins, Loretta Jones, 2007-08. Written for calculus inclusive general chemistry courses. Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications. Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems, think about nature and matter and visualize chemical concepts as working chemists do. Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text. Instructors have the option of whether to incorporate calculus in the coverage of topics. The multimedia integration of Chemical Principles is more deeply established than any other text for this course. Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding.

Cracking the AP Chemistry Exam, 2016 Edition Princeton Review, 2015-08-11. EVERYTHING YOU NEED TO HELP SCORE A PERFECT 5. Equip yourself to ace the AP Chemistry Exam with The Princeton Review's comprehensive study guide including 2 full length practice tests thorough content reviews access to our AP Connect online portal and targeted strategies for every section of the exam. This eBook edition is optimized for on screen learning with cross linked questions answers and explanations. We don't have to tell you how tough AP Chem is or how important a stellar score on the AP exam can be to your chances of getting into a top college of your choice. Written by Princeton Review experts who know their way around chem. Cracking the AP Chemistry Exam will give you the help you need to get the score you want. Techniques That Actually Work. Tried and true strategies to help you avoid traps and beat the test. Tips for pacing yourself and guessing logically. Essential tactics to help you work smarter not

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Review of Exchange Arrangements, Restrictions, and Controls International Monetary Fund. Monetary and Capital Markets Department, 2007-11-27 This review reports on trends developments and issues in exchange rate arrangements and currency convertibility This section presents a summary of the overall findings Section II provides an overview of key trends and developments in exchange rate arrangements Section III outlines key trends and developments in current and capital account restrictions The present paper uses the existing methodology for the classification of exchange rate arrangements

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Study Guide to accompany Basic Concepts of Chemistry, 7th Edition Leo J. Malone, 2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry

McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill s SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need perform at your best This book includes 4

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Mastering NEET MCQs: Speed-solving Techniques for Medical Aspirants Dr. Ashokan Kannarath,2024-01-31 This book dives deep into the strategies techniques and tips required to excel in the NEET MCQs It is designed to help you not only solve questions quickly but also enhance your understanding and retention of vital medical concepts It explains how to analyse NEET MCQs without getting overwhelmed efficiently identifying keywords understanding question patterns and eliminating wrong answer options It also exposes proven strategies to increase your answering speed without compromising accuracy Includes time management tips mental math tricks and effective test taking methods

Strengthening the ILO's Capacity to Assist Its Members' Efforts to Reach Its Objectives in the Context of Globalization ,2007 Examines how the ILO's capacity could be strengthened to fulfil its mandate in the context of globalization Reflects on how changes in governance structures could help the Organization over the medium term to align more closely the needs of constituents assessments of the best ways to meet those needs the development of effective services by the Office and the scope for mobilizing technical and other resources

Safeguards Assessments - 2025 Update International Monetary Fund. Finance Dept.,2025-09-30 Safeguards assessments are a key pillar of the IMF's risk management framework They aim to mitigate the risks of misuse of Fund resources and misreporting of program monetary data These assessments are mandatory for Fund supported financing arrangements and are complemented by ongoing monitoring for as long as credit remains outstanding While the full safeguards assessment reports are confidential the IMF Executive Board receives a summary of safeguards findings and recommendations through member country staff reports In addition the Board also receives periodic reports on safeguards activities This period covers safeguards activity between May 2022 and April 2025

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change

Kaplan SAT: Chemistry Claire Aldridge,Karl Lee,2006-02 Includes 3 Full length practice tests Comprehensive review of all the tested material with practice quizzes for each chapter Effective strategies to maximize your score

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