

PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



Atomic symbol: H
Atomic number: 1
Atomic weight: 1.0079

Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point: -259.14 °C (14.009985 K, -434.45203 °F)

Boiling Point: -252.87 °C (20.280005 K, -423.166 °F)

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293 K: 0.08988 g/cm³

Color: colorless

Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

Isotopes

Isotope	Half Life
H-1	Stable
H-2	Stable
H-3	12.3 years

Facts

Name Origin: Greek words *hudeir* (water) and *gennan* (generate)

Uses: Balloons, metal refining

Obtained From: mines, oil, gas wells

Molecular Structure of Carbon



Atomic symbol: C
Atomic number: 6
Atomic weight: 12.011

Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0107 amu

Melting Point: 3500.0 °C (3773.15 K, 6332.0 °F)

Boiling Point: 4827.0 °C (5100.15 K, 8720.6 °F)

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293 K: 2.62 g/cm³

Color: May be black

Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2

Second Energy Level: 4

Isotopes

Isotope	Half Life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.5 seconds

Facts

Uses: steel, filters

Obtained From: burning with insufficient oxygen

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plato again assume x where a and b are integers and $b \neq 0$

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