

CH 5.3 - Periodic Trends WebQuest PART I

There is a .pdf version of this assignment on my blog. Please utilize the online version to link the webpages in order to avoid typing in the addresses. Reminder: columns on the periodic table are called families or groups and rows are called periods

Atomic Radius (pages 174-175)

Using your text, please answer the following:

1. Can the radius of an atom be measured directly? _____
2. In what unit is atomic radius usually measured? _____
3. Define Atomic Radius: _____

Go to the following website:

http://www.webelements.com/periodicity/atomic_radius_empirical/bar_chart.html

4. Look at the graph. What is on the X Axis? _____
5. What is on the Y Axis (include units)? _____
6. **The colors on the graph represent rows on the periodic table.**
In a complete sentence, answer - What appears to be the trend in atomic radius as you move from left to right in a row?

In a complete sentence, answer - What appears to be the trend in atomic radius as you move down a column?

7. Using your text, please answer the following:

- a. What is the general trend for atomic size within a group/family from top to bottom?
- b. What two variables affect atomic size within a group?
- c. What is the general trend in atomic size from left to right within a period?

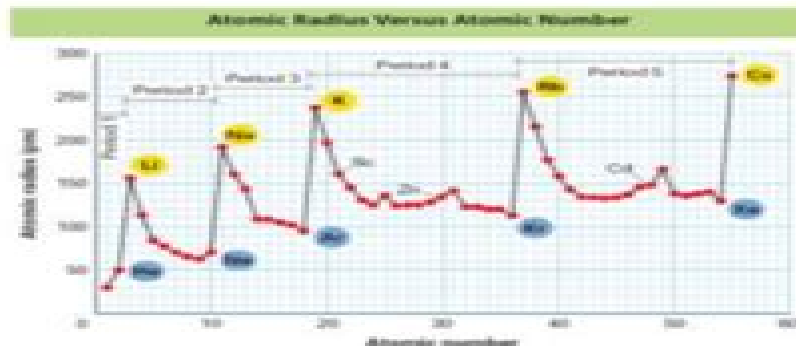
8. Use the graph to the right to answer the following:

- a. Determine which metal has an atomic radius of 238 pm?
- b. Is an atom of barium, atomic number 56, smaller or larger than an atom of cesium (Cs)?
- c. Which has a larger atom, helium or argon?
- d. Which has a larger atom, potassium or argon?

Teacher Checkpoint _____

Ions and Ionization Energy (pages 177-181)

Using your text, please answer the following:



Periodic Trends Webquest Answers

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Periodic Trends Webquest Answers:

Periodic Trends Webquest Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Trends Webquest Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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