

| Group name*                  |    |               |
|------------------------------|----|---------------|
| Hydrogen & the alkali metals | 14 | Crystallogens |
| Alkaline earth metals        | 15 | Psithogens    |
| Coinage metals               | 16 | Chalcogens    |
| Volatile metals              | 17 | Halogens      |
| Isotopans                    | 18 | Helium atom   |

\*Groups 1 to 20 are ranked after their first matches. i.e. Group 1 is the top-ranked group.

|    |    |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     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| 87 | 88 | 89 | 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 | 104 | 105 | 106 | 107 | 108 | 109 | 110 | 111 | 112 | 113 | 114 | 115 | 116 | 117 | 118 | 119 | 120 | 121 | 122 | 123 | 124 | 125 | 126 | 127 | 128 | 129 | 130 | 131 | 132 | 133 | 134 | 135 | 136 | 137 | 138 | 139 | 140 | 141 | 142 | 143 | 144 | 145 | 146 | 147 | 148 | 149 | 150 | 151 | 152 | 153 | 154 | 155 | 156 | 157 | 158 | 159 | 160 | 161 | 162 | 163 | 164 | 165 | 166 | 167 | 168 | 169 | 170 | 171 | 172 | 173 | 174 | 175 | 176 | 177 | 178 | 179 | 180 | 181 | 182 | 183 | 184 | 185 | 186 | 187 | 188 | 189 | 190 | 191 | 192 | 193 | 194 | 195 | 196 | 197 | 198 | 199 | 200 | 201 | 202 | 203 | 204 | 205 | 206 | 207 | 208 | 209 | 210 | 211 | 212 | 213 | 214 | 215 | 216 | 217 | 218 | 219 | 220 | 221 | 222 | 223 | 224 | 225 | 226 | 227 | 228 | 229 | 230 | 231 | 232 | 233 | 234 | 235 | 236 | 237 | 238 | 239 | 240 | 241 | 242 | 243 | 244 | 245 | 246 | 247 | 248 | 249 | 250 | 251 | 252 | 253 | 254 | 255 | 256 | 257 | 258 | 259 | 260 | 261 | 262 | 263 | 264 | 265 | 266 | 267 | 268 | 269 | 270 | 271 | 272 | 273 | 274 | 275 | 276 | 277 | 278 | 279 | 280 | 281 | 282 | 283 | 284 | 285 | 286 | 287 | 288 | 289 | 290 | 291 | 292 | 293 | 294 | 295 | 296 | 297 | 298 | 299 | 300 | 301 | 302 | 303 | 304 | 305 | 306 | 307 | 308 | 309 | 310 | 311 | 312 | 313 | 314 | 315 | 316 | 317 | 318 | 319 | 320 | 321 | 322 | 323 | 324 | 325 | 326 | 327 | 328 | 329 | 330 | 331 | 332 | 333 | 334 | 335 | 336 | 337 | 338 | 339 | 340 | 341 | 342 | 343 | 344 | 345 | 346 | 347 | 348 | 349 | 350 | 351 | 352 | 353 | 354 | 355 | 356 | 357 | 358 | 359 | 360 | 361 | 362 | 363 | 364 | 365 | 366 | 367 | 368 | 369 | 370 | 371 | 372 | 373 | 374 | 375 | 376 | 377 | 378 | 379 | 380 | 381 | 382 | 383 | 384 | 385 | 386 | 387 | 388 | 389 | 390 | 391 | 392 | 393 | 394 | 395 | 396 | 397 | 398 | 399 | 400 | 401 | 402 | 403 | 404 | 405 | 406 | 407 | 408 | 409 | 410 | 411 | 412 | 413 | 414 | 415 | 416 | 417 | 418 | 419 | 420 | 421 | 422 | 423 | 424 | 425 | 426 | 427 | 428 | 429 | 430 | 431 | 432 | 433 | 434 | 435 | 436 | 437 | 438 | 439 | 440 | 441 | 442 | 443 | 444 | 445 | 446 | 447 | 448 | 449 | 450 | 451 | 452 | 453 | 454 | 455 | 456 | 457 | 458 | 459 | 460 | 461 | 462 | 463 | 464 | 465 | 466 | 467 | 468 | 469 | 470 | 471 | 472 | 473 | 474 | 475 | 476 | 477 | 478 | 479 | 480 | 481 | 482 | 483 | 484 | 485 | 486 | 487 | 488 | 489 | 490 | 491 | 492 | 493 | 494 | 495 | 496 | 497 | 498 | 499 | 500 | 501 | 502 | 503 | 504 | 505 | 506 | 507 | 508 | 509 | 510 | 511 | 512 | 513 | 514 | 515 | 516 | 517 | 518 | 519 | 520 | 521 | 522 | 523 | 524 | 525 | 526 | 527 | 528 | 529 | 530 | 531 | 532 | 533 | 534 | 535 | 536 | 537 | 538 | 539 | 540 | 541 | 542 | 543 | 544 | 545 | 546 | 547 | 548 | 549 | 550 | 551 | 552 | 553 | 554 | 555 | 556 | 557 | 558 | 559 | 560 | 561 | 562 | 563 | 564 | 565 | 566 | 567 | 568 | 569 | 570 | 571 | 572 | 573 | 574 | 575 | 576 | 577 | 578 | 579 | 580 | 581 | 582 | 583 | 584 | 585 | 586 | 587 | 588 | 589 | 590 | 591 | 592 | 593 | 594 | 595 | 596 | 597 | 598 | 599 | 600 | 601 | 602 | 603 | 604 | 605 | 606 | 607 | 608 | 609 | 610 | 611 | 612 | 613 | 614 | 615 | 616 | 617 | 618 | 619 | 620 | 621 | 622 | 623 | 624 | 625 | 626 | 627 | 628 | 629 | 630 | 631 | 632 | 633 | 634 | 635 | 636 | 637 | 638 | 639 | 640 | 641 | 642 | 643 | 644 | 645 | 646 | 647 | 648 | 649 | 650 | 651 | 652 | 653 | 654 | 655 | 656 | 657 | 658 | 659 | 660 | 661 | 662 | 663 | 664 | 665 | 666 | 667 | 668 | 669 | 670 | 671 | 672 | 673 | 674 | 675 | 676 | 677 | 678 | 679 | 680 | 681 | 682 | 683 | 684 | 685 | 686 | 687 | 688 | 689 | 690 | 691 | 692 | 693 | 694 | 695 | 696 | 697 | 698 | 699 | 700 | 701 | 702 | 703 | 704 | 705 | 706 | 707 | 708 | 709 | 710 | 711 | 712 | 713 | 714 | 715 | 716 | 717 | 718 | 719 | 720 | 721 | 722 | 723 | 724 | 725 | 726 | 727 | 728 | 729 | 730 | 731 | 732 | 733 | 734 | 735 | 736 | 737 | 738 | 739 | 740 | 741 | 742 | 743 | 744 | 745 | 746 | 747 | 748 | 749 | 750 | 751 | 752 | 753 | 754 | 755 | 756 | 757 | 758 | 759 | 760 | 761 | 762 | 763 | 764 | 765 | 766 | 767 | 768 | 769 | 770 | 771 | 772 | 773 | 774 | 775 | 776 | 777 | 778 | 779 | 780 | 781 | 782 | 783 | 784 | 785 | 786 | 787 | 788 | 789 | 790 | 791 | 792 | 793 | 794 | 795 | 796 | 797 | 798 | 799 | 800 | 801 | 802 | 803 | 804 | 805 | 806 | 807 | 808 | 809 | 810 | 811 | 812 | 813 | 814 | 815 | 816 | 817 | 818 | 819 | 820 | 821 | 822 | 823 | 824 | 825 | 826 | 827 | 828 | 829 | 830 | 831 | 832 | 833 | 834 | 835 | 836 | 837 | 838 | 839 | 840 | 841 | 842 | 843 | 844 | 845 | 846 | 847 | 848 | 849 | 850 | 851 | 852 | 853 | 854 | 855 | 856 | 857 | 858 | 859 | 860 | 861 | 862 | 863 | 864 | 865 | 866 | 867 | 868 | 869 | 870 | 871 | 872 | 873 | 874 | 875 | 876 | 877 | 878 | 879 | 880 | 881 | 882 | 883 | 884 | 885 | 886 | 887 | 888 | 889 | 890 | 891 | 892 | 893 | 894 | 895 | 896 | 897 | 898 | 899 | 900 | 901 | 902 | 903 | 904 | 905 | 906 | 907 | 908 | 909 | 910 | 911 | 912 | 913 | 914 | 915 | 916 | 917 | 918 | 919 | 920 | 921 | 922 | 923 | 924 | 925 | 926 | 927 | 928 | 929 | 930 | 931 | 932 | 933 | 934 | 935 | 936 | 937 | 938 | 939 | 940 | 941 | 942 | 943 | 944 | 945 | 946 | 947 | 948 | 949 | 950 | 951 | 952 | 953 | 954 | 955 | 956 | 957 | 958 | 959 | 960 | 961 | 962 | 963 | 964 | 965 | 966 | 967 | 968 | 969 | 970 | 971 | 972 | 973 | 974 | 975 | 976 | 977 | 978 | 979 | 980 | 981 | 982 | 983 | 984 | 985 | 986 | 987 | 988 | 989 | 990 | 991 | 992 | 993 | 994 | 995 | 996 | 997 | 998 | 999 | 1000 |
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☐ **Radioactive**  
 Element is  
 likely to be either  
 a metalloid or a  
 good transition  
 metal

1000



Moderate to high density metals, with high melting and boiling points, many have high hardness and mechanical strength. Those from R1 onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.



Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalytic properties. Very easily electronegative; insulating oxides.

**Figure 1**

Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater simplicity or fewer nearest neighbours than other metals. All used or treated as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterised, to varying degrees, by covalent bonding tendencies, anisotropies, and ionic species such as  $\text{Zn}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Si}^{4+}$ , and  $\text{Sn}^{4+}$ .

$\text{N}_2$ ,  $\text{O}_2$ , and  $\text{Cl}_2$  are, respectively, colorless, pale yellow, and pale yellow-green gases;  $\text{Br}_2$  is a red-brown liquid;  $\text{I}_2$  is a metallic looking solid. Chemically, iodine has to have high ionisation energies, electron affinities, and electronegativity values, and be relatively strong oxidising agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically corrosive nature of the non-metallic halogens. All four tend to form predominantly ionic compounds with metals. Their oxides are acidic.

Colorless, odorless, non-flammable gases with very low chemical reactivity. The first noble gas compound, of approximate composition  $\text{XePtF}_6$ , was not prepared until 1962. More than half-a-thousand noble gas compounds were synthesized in the ensuing half-century. Compounds of He, Ne and Ar, in ambient conditions, are not known.

Then start as for  $H$ , and then follow up with a Lie bracket like  $[X_1, X_2]$  and also  $[X_1, X_3]$ . I don't think you need to do  $X_1$  and the

- *Go forward* means you move forward through the groups (1-2)
- *Go backward* means you move backward through the groups (2-1)

# Periodic Table Unscrambled Table Answers

**Chris McMullen, Carolyn Kivett**



## Periodic Table Unscrambled Table Answers:

*Microsoft Word Simple Projects* Jan Rader, Jan Ray, 2001-04 Projects for language arts social studies science and math Provided templates can be modified to meet specific needs Project samples also provided *Periodic Table* Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket **Scientifica**, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **Unscramble Your Nest Egg** John J. Cunningham, 2013-10-25 Pay off mortgage by age 50 55 deposit former mortgage payments into income earning accounts and result will be a cash nest egg big enough to pay 50 percent of living expenses during 20 years of retirement New Modular Science for GCSE. Ann Fullick, 1998 This homework book provides graded homework questions to help the teacher set homework for a range of abilities It offers 14 sets of homework questions for each of the 12 modules in the NEAB science syllabus **Scientifica Pupil Book 8 (Levels 4-7)** David Sang, Lawrie Ryan, Jane Taylor, Peter Ellis, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **The Science Teacher**, 2008 **Pathways in Science: Chemistry (Pt. 1-3)** Joseph M. Oxenhorn, 1970 **Cyber Science 6 Tm' 2007 Ed.**, **Story of the World Activity Book 4 Modern Age** Susan Wise Bauer, 2005-11-29 Presents a history of the ancient world from 6000 B C to 400 A D Amazing People: Scientists and Mathematicians Elise Craver, Malaski, Malay, Hailey Scragg, 2023-01-23 Inspirational 1st 2nd and 3rd grade workbooks for kids ages 7 Motivate and inspire children with the Amazing People Scientists and Mathematicians Activity Book 1st 2nd and 3rd grade books are a great way for children to learn about inspirational people who have made important contributions to science and math through a variety of activities that are both fun AND educational Why You ll Love This 1st Grade Third Grade Workbook Engaging and educational activities and wall d cor The activity book features biographies activities and flash cards covering 53 inspirational mathematicians and scientists The activity book also includes 1 inspirational poster great for both homeschool and classroom curriculum and classroom d cor Tracking progress along the way Test your child s knowledge with the flash cards that cover scientists and mathematicians learned throughout the book Use the stickers to reward students on a job well done after completing activities Practically sized for every activity

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**Verbal Reactions - Word Scrambles With a Chemical Flavor (Easy)** Chris McMullen,Carolyn Kivett,2011-12-09 VERBAL ReACTiONS are a new form of word scrambles with a chemical flavor You don t need to know any science to be able to solve VERBAL ReACTiONS puzzles but the puzzles bear a resemblance to chemical reactions EXAMPLE Here is a sample Es 2 S P Si 2 O N \_ \_ \_ \_ \_ \_ \_ \_ This VERBAL ReACTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that s why there are 8 blanks in the puzzle ANSWER For this puzzle the answer is P O S S Es Si O N possession These VERBAL ReACTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VERBAL ReACTiON by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using symbols from the periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Easy volume consists of words with 4 to 5 symbols which involves familiarity with common 4 to 10 letter words Other Medium and Hard volumes consist of longer words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U \_ \_ \_ \_ \_ 2 2 C N 2 I P \_ \_ \_ \_ \_ 3 Ti C Cr P Y \_ \_ \_ \_ \_ 4 2 C U 2 S Es \_ \_ \_ \_ \_ You can find the answers below Note that this easy volume consists of chemical words with 4 to 5 symbols We recommend starting with our easy puzzles before tackling the medium or hard puzzles available in separate volumes

ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS

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*Verbal Reactions - Word Scrambles With a Chemical Flavor (Hard)* Chris McMullen, Carolyn Kivett, 2011-06-18

VERBAL REACTIONS are a new form of word scrambles with a chemical flavor You don't need to know any science to be able to solve VERBAL REACTIONS puzzles but the puzzles bear a resemblance to chemical reactions Here is a sample  $Es_2 S P Si_2 O N \underline{\hspace{1cm}}$  This VERBAL REACTION is a word scramble consisting of one Es two Ss one P one Si two Os and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that's why there are 8 blanks in the puzzle For this puzzle the answer is POSSESION These VERBAL REACTIONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VERBAL REACTION by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using symbols from the periodic table For example the chemical word POSSESION is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don't need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Hard volume consists of words with 7 to 8 symbols which involves familiarity with common 8 to 14 letter words Other Medium and Easy volumes consist of shorter words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution

MORE EXAMPLES  $1 S Ni Ge U \underline{\hspace{1cm}}$   $2 2 C N 2 I P \underline{\hspace{1cm}}$   $3 Ti C Cr P Y \underline{\hspace{1cm}}$   $4 2 C U 2 S Es \underline{\hspace{1cm}}$  You can find the answers at the end of this paragraph

Note that this hard volume consists of chemical words with 7 to 8 symbols which are longer than the examples shown here We recommend starting with our easy or medium puzzles before tackling these hard puzzles available in separate volumes

ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS

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