

SDSU CHEMISTRY PLACEMENT TEST WITH COMPLETE SOLUTIONS 100% 2023/2024

proton
-heavy
-positively charged
-nucleon

neutron
-heavy
-neutral
-nucleon

electron
-very small (1/1800 of proton)
-negative
-outside of nucleus

properties of nucleus
-# of protons determines the element
-held together by nuclear force
-# of protons is extremely stable unlike # of electrons

atomic number
equal to the # of protons

atomic mass
of protons + # of neutrons averaged across all of the element on earth

isotopes
-when elements have different masses but the same chemical properties
-ex. silver-107 and silver-109

lewis dot
-a 2-D model that represents covalent bonds as straight lines, ionic bonds as spaces, and unbonded valence electrons as dots
-can have single, double, or triple bonds

physical properties of compounds and molecules
-melting point
-boiling point
-solubility
-electrical conductivity

molecular mass
of atoms(atomic mass)=molecular mass

Sdsu Chemistry Placement Test Study Guide

Julia Schneider



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