



# Samsung Dsb B350v Manual

**Liying Dong**



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the aldol condensation is extremely important because it can form a  $\beta$  hydroxy aldehyde or

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104 110 c this helped conclude that dibenzalacetone had a trans trans e e isomer

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produce an  $\alpha,\beta$  unsaturated aldehyde and water figure 1 aldol condensation of ethanal to