

ICSE Class 10 Chemistry Question Paper Solution 2016

CHEMISTRY SCIENCE Paper – 2

Question 1

- (a) Fill in the blanks with the choices given in brackets. [5]
- (i) Metals are good _____ (*oxidizing agents / reducing agents*) because they are electron _____ (*acceptors/ donors*).
 - (ii) Electrovalent compounds have _____ (*high / low*) melting points.
 - (iii) Higher the pH value of a solution, the more _____ (*acidic / alkaline*) it is.
 - (iv) _____ (*AgCl / PbCl₂*), a white precipitate is soluble in excess NH_4OH .
 - (v) Conversion of ethene to ethane is an example of _____ (*hydration / hydrogenation*).
- (b) Choose the *correct answer* from the options given below: [5]
- (i) An element with the atomic number 19 will most likely combine chemically with the element whose atomic number is:
 - A. 17
 - B. 11
 - C. 18
 - D. 20
 - (ii) The ratio between the number of molecules in 2g of hydrogen and 32g of oxygen is:
 - A. 1:2
 - B. 1:0.01
 - C. 1:1
 - D. 0.01:1 [Given that H=1, O=16]
 - (iii) The two main metals in *Bronze* are:
 - A. Copper and zinc
 - B. Copper and lead
 - C. Copper and nickel
 - D. Copper and tin
 - (iv) The particles present in *strong electrolytes* are:
 - A. only molecules
 - B. mainly ions

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Table of Contents Sample Question Papers Solutions

1. Understanding the eBook Sample Question Papers Solutions
 - The Rise of Digital Reading Sample Question Papers Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Question Papers Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Question Papers Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Question Papers Solutions
 - Personalized Recommendations
 - Sample Question Papers Solutions User Reviews and Ratings
 - Sample Question Papers Solutions and Bestseller Lists
5. Accessing Sample Question Papers Solutions Free and Paid eBooks
 - Sample Question Papers Solutions Public Domain eBooks
 - Sample Question Papers Solutions eBook Subscription Services
 - Sample Question Papers Solutions Budget-Friendly Options

6. Navigating Sample Question Papers Solutions eBook Formats
 - ePub, PDF, MOBI, and More
 - Sample Question Papers Solutions Compatibility with Devices
 - Sample Question Papers Solutions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Question Papers Solutions
 - Highlighting and Note-Taking Sample Question Papers Solutions
 - Interactive Elements Sample Question Papers Solutions
8. Staying Engaged with Sample Question Papers Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Question Papers Solutions
9. Balancing eBooks and Physical Books Sample Question Papers Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Question Papers Solutions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sample Question Papers Solutions
 - Setting Reading Goals Sample Question Papers Solutions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sample Question Papers Solutions
 - Fact-Checking eBook Content of Sample Question Papers Solutions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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