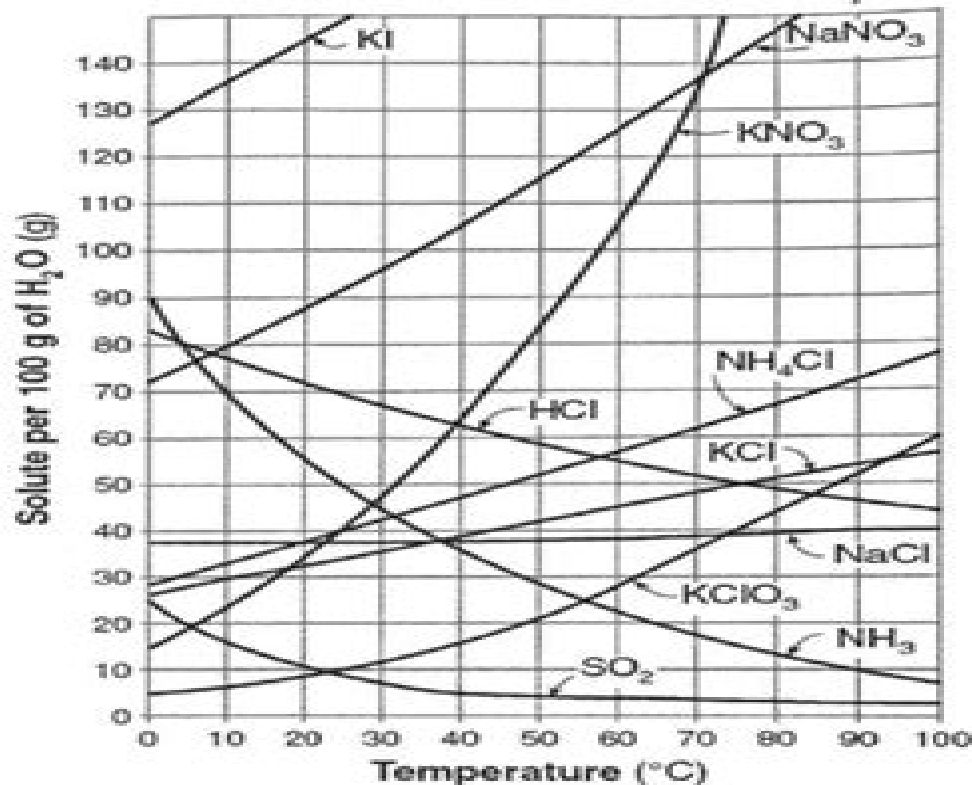


Reading a Solubility Chart

- 1) The curve shows the # of grams of solute in a saturated solution containing 100 mL or 100 g of water at a certain temperature.
- 2) Any amount of solute below the line indicates the solution is unsaturated at a certain temperature
- 3) Any amount of solute above the line in which all of the solute has dissolved shows the solution is supersaturated.
- 4) If the amount of solute is above the line but has not all dissolved, the solution is saturated and the # grams of solute settled on the bottom of the container = total # g in solution - # g of a saturated solution at that temperature. (according to the curve)
- 5) Solutes whose curves move upward w/ increased temperature are typically solids b/c the solubility of solids increases w/ increased temperature.
- 6) Solutes whose curves move downward w/ increased temperature are typically gases b/c the solubility of gases decreases with increased temperature.



Solubility Problems

1. Which of the salts shown on the graph is the most soluble in water at 10°C? **KI**
2. At what temperature is 90g of NaNO₃ a saturated solution in 100grams of water? **≈ 23°C**
3. At solution has 70grams of KNO₃ and is unsaturated, what is the minimum temperature? **46°C**
4. At 20°C, a saturated solution of KClO₃ is created in water. How many grams of potassium chlorate must be added to saturate the solution at 50°C? **12g**
5. Which of the salts shown on the graph is the least soluble in water at 10°C? **KClO₃**
6. Which of the salts shown on the graph has the greatest increase in solubility as the temperature increases from 30 degrees to 60 degrees? **KNO₃**

Reading The Solubility Chart Problems Answers With Work

Victor M. Corman



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Popular Science Monthly ,1927 Biochemistry of Silicon and Related Problems Gerd Bendz,2013-06-29 Silicon chemistry was initiated in 1823 by Berzelius who prepared elemental silicon In many ways silicon was considered a typical opposite of carbon although the two elements are closely related as to their electronic structure both having four valence electrons The properties of their compounds are however extreme ly different Both form extended structures but in different ways carbon by covalent carbon carbon bonds silicon by polar silicon oxygen silicon bonds The complex carbon compounds are integral parts of all living matter plants and animals The corresponding silicon compounds build up a major part of dead matter soils and minerals As recently as twenty years ago the title of this Symposium BiOChemistry of Silicon would have been considered as contradictio in adjecto However the development in the field has during the past fifteen years been overwhelming and has convinced us that silicon is a necessary element in the life processes for animals as well as for plants Interesting therapeutical uses have been suggested but we have also become increasingly aware of serious occupational diseases asbestosis and silicosis and of possible cancerogenic effects It is our hope that this volume will give some idea about various aspects of silicon compounds which were discussed during the Symposium

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Chemistry in the Community. American

Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts Nature, 1917 The Journal of the Chemical, Metallurgical & Mining Society of South Africa, 1914 *Handbook of Adolescent Literacy Research* Leila Christenbury, Randy Bomer, Peter Smagorinsky, 2011-06-10 The first comprehensive research handbook of its kind this volume showcases innovative approaches to understanding adolescent literacy learning in a variety of settings Distinguished contributors examine how well adolescents are served by current instructional practices and highlight ways to translate research findings more effectively into sound teaching and policymaking The book explores social and cultural factors in adolescents approach to communication and response to instruction and sections address literacy both in and out of schools including literacy expectations in the contemporary workplace Detailed attention is given to issues of diversity and individual differences among learners Winner Literacy Research Association's Fry Book Award *Coke & Chemistry, U.S.S.R.*, 1973

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