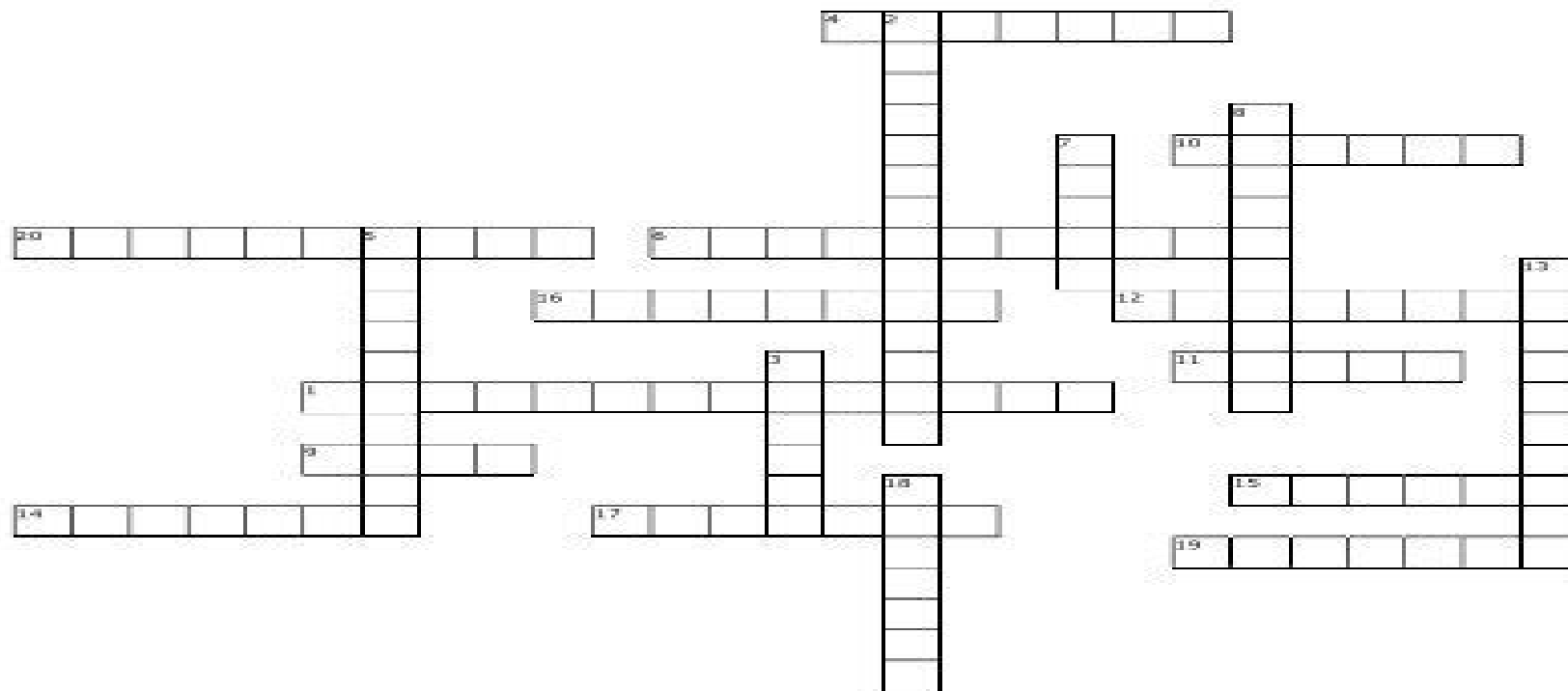


Periodic Table Puzzle



Across:

1. Groups 1, 2, and 13-18 (or 1A-8A) are referred to as the ____ elements
4. The name of the group 17 elements is the ____ group
6. The name of the group 18 elements is the ____
9. The ____ number identifies the total number of protons and neutrons in the atom
10. The columns in the periodic table are called ____
11. The majority of the elements in the periodic table exist in this state at room temperature
12. The ____ Series consists of elements whose atomic numbers are 89-103
14. The rows in the periodic table are called ____
15. The name of the Group 1 elements is the ____ metals
16. The only non-metal on the left side of the periodic table
17. This element is the only liquid metal at room temperature
19. The part of the atom that contains the protons and the neutrons
20. The ____ Series consists of elements whose atomic numbers are 57-71

Down:

2. The name of the group 2 elements is the ____ Metals
3. The ____ number identifies the number of protons in an atom
5. These elements are not shiny at all
7. This type of element can conduct heat and electricity
8. Groups 3-12 (or 1B-8B) are referred to as the ____ elements
13. These elements exhibit some properties of metals and some properties of non-metals
18. This element is the only liquid non-metal at room temperature

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and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives The Elements Tom Jackson,Worth Press Ltd,2012-01-01 In this essential guide to the Periodic Table we track the history of the powerful yet elegant tool that lays bare the building blocks of the Universe The journey begins just as the first cities are forming and follows the contributions made by philosophers alchemists industrialists and great scientists as they gather force to create this masterpiece of accumulated knowledge Includes a removable fold out concertina neatly housed in the back of the book This fold out provides a 12 page Timeline History of the Periodic Table that embeds the story of chemistry in historical context and shows Who Did What When at a glance *Mastering the Periodic Table* Linda Trombley,Faye Williams,2000 Whether students are studying chemistry biology or other sciences the periodic table is a vitally important tool These 50 word games puzzles and other creative activities unlock the nature of the various elements while explicating periodicity atomic structure element groups and more Complete teacher support includes background information answer keys and materials lists **Leveled Texts: The Periodic Table** Joshua BishopRoby,2014-01-01 All students can learn about the periodic table through text written at four different reading levels Symbols on the pages represent reading level ranges to help differentiate instruction Provided comprehension questions complement the text **The Basics of the Periodic Table** Leon Gray,2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table Included are discussions of the discovery of matter atoms atomic structure molecules compounds ions and isotopes as well as the first identifications of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events Mystery of the Periodic Table Benjamin D Wiker,2003-04-18 Leads the reader on a delightful and absorbing journey through the ages on the trail of the elements of the Periodic Table as we know them today He introduces the young reader to people like Von Helmont Boyle Stahl Priestly Cavendish Lavoisier and many others all incredibly diverse in personality and approach who have laid the groundwork for a search that is still unfolding to this day The first part of Wiker s witty and solidly instructive presentation is most suitable to middle school age while the later chapters are designed for ages

12 13 and up with a final chapter somewhat more advanced Illustrated by Jeanne Bendick and Ted Schluenderfritz *The periodic table* Brian J. Knapp,1998 Twelve Volume Reference for beginning chemistry students **Decoding the Periodic Table** Jurjen Van Van der Wal,2010-06-16 Decoding the Periodic Table In the year 1809 Russian Chemist Dimitri Ivanovich Mendeleev published the first Chemical Periodic Table of the Elements Since then scientists developed a theory that every element consists of a tightly packed mass of protons and neutrons which is surrounded by concentric spheres which are occupied by orbiting electrons This concept is known as the Standard Model but it is flawed because it cannot accommodate gravity For many decades researchers including this author in the late 1960 s tried to formulate theories that in a comprehensive way would unite all physical designs and phenomena under a common denominator but to no avail The String Theory is one of them but it has found no success In 1989 the author of the book made a fresh start again with the notion that particles such as the proton neutron and electron would not be round but that they would have straight lined square or cubic configurations because the Periodic Table s had quadratic numbers in their quantities of elements such as Period 1 with 2 elements 2 2x12 Periods 2 3 8 2 x 22 Periods 4 5 18 2 x 32 and Period 6 7 32 2 x 42 This might indicate square structural relationships lets give it a try It worked it took 20 years to do it it was a fascinating journey falling down and getting up again and again working without a computer with all figures hand drawn and discovering that the forces that make the nuclear mass of the neutron do that by multiplying two force filled cubic volumes as in $3F \cdot 6 \cdot 3F \times 3F \times 3 \times 3F \times 3F \times 3F$ 1838 2656 *Periodic Table Basic* Mark Jackson,2002-07-16 Periodic table of the elements **Quiz Yourself Clever!** **The Periodic Table** DK,2024-10-03 Get to grips with the periodic table with this fact packed quiz book Learn all the ingredients that make up the universe in this quiz book for children aged 9 that breaks down the periodic table Quiz Yourself Clever The Periodic Table goes through all 118 elements in an accessible and kid friendly way showing the pure form of every element through an eye catching image surrounded by quiz style questions This periodic table quiz book for children offers Fast and fun learning through the use of detailed images and interesting facts Hundreds of quick fire questions for children to quiz themselves and challenge friends and family Information that has been endorsed and authenticated by experts Vivid pictures and fast facts will allow children to gather a wealth of knowledge in a fun way Turn the page to find the answers contained in a handy dataset from its atomic number and mass to its melting point and when it was discovered Images of how it appears in nature and how we use it give a fuller understanding of each element More in the series If you enjoyed Quiz Yourself Clever The Periodic Table then why not test yourself other quiz books to boost your knowledge like Quiz Yourself Clever Rocks Minerals to get to grips with the fascinating rock and mineral forms all over the world or Animals of the World to learn about the most fascinating wildlife on our planet

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