

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)**1. Physical Science Basics**

- a. What tool(s) would you want to use to find the:
- i. Mass of an object? triple beam balance Basic SI Unit? kg
 - ii. Volume of an object? ruler or water displacement method Basic SI Unit? cm^3 or ml
 - iii. Length of an object? Ruler Basic SI Unit? meter
 - iv. Density of an object? graduated cylinder Basic SI Unit? ml
- b. What is the formula for calculating density? mass/volume
- i. What is the density of a cube that is 25 g and has a volume of 3 cm^3 ? $8\frac{1}{3}\text{g/cm}^3$
 - ii. Compare the density of solid H_2O to liquid H_2O , and explain what observable effect this has.
solid H_2O less dense than liquid H_2O

2. Chemistry Basics

- a. Element (def.) substance made up of one kind of atom
- b. Atom (def.) basic building block of matter
- c. Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- d. Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- a. The current periodic table is arranged by increasing atomic number.
- b. Reading the Periodic Table
- i. An element's properties can be predicted from its location in the periodic table.
 - ii. The atomic number represents the number of protons.
 - iii. The atomic mass represents the number of neutrons and protons.

Physical Science Final Exam Study Guide

Kevin Smith



Physical Science Final Exam Study Guide:

X-kit Exam 2004 Physical Science, 2005 **Physics Handbook & Study Guide Grade 11-12 IEB** Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text CHEMISTRY HANDBOOK & STUDY GUIDE Gr11-12 NE Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text **Resources in Education**, 1997

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AQA Mathematics Bob Hartman, 2006 Written by examiners and practising teachers this work offers study and homework support throughout GCSE It is useful as a reference source a

lesson back up and a revision guide *Research in Education* ,1969-05 *The Design and Layout of Fire Sprinkler Systems*
 Mark Bromann,2001-04-05 Although effective fire sprinkler systems are crucial to public safety for years the designers of those systems had few published resources to reference and guide them through their design processes The first edition of this book changed all that and now *The Design and Layout of Fire Sprinkler Systems* Second Edition suits their needs even better Written and thoroughly updated by a fire prevention engineer with more than 20 years of experience this book provides a complete systematic introduction to automatic fire sprinkler design and layout from design basics code requirements and pipe hanging to hydraulic calculations retrofits and details on fire pumps The author carefully outlines all of a designer s responsibilities and includes an entire chapter dedicated to preparing for the NICET exam More than 150 sample diagrams checklists sample forms spec sheets photographs and a glossary complement the text and the larger page size of this edition permits clear presentation of diagrams and schematics *The Design and Layout of Fire Sprinkler Systems* not only builds the foundation and skills of newcomers to the field but also provides an outstanding reference for fire safety professionals building inspectors insurance underwriters and municipal officials **Departments of Veterans Affairs and Housing and Urban Development and Independent Agencies Appropriations for Fiscal Year 1993: American Battle Monuments Commission** United States. Congress. Senate. Committee on Appropriations. Subcommittee on VA-HUD-Independent Agencies,1992 **CLEP® Natural Sciences Book + Online** Laurie Ann Callihan, Ph.D,2016-06-08
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Resources for Teaching Middle School

Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents

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