

## Chapter 6 Chemical Bonds

**Section 6.2 Covalent Bonding****(pages 165–169)**

*This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.*

**Reading Strategy (page 165)**

**Relating Text and Visuals** As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

**Covalent Bonds (pages 165–167)**

- Describe a covalent bond. \_\_\_\_\_  
\_\_\_\_\_
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - electron dot
  - structural formula
  - space-filling
  - electron cloud
- Describe a molecule. \_\_\_\_\_  
\_\_\_\_\_
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. \_\_\_\_\_
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - zero
  - one
  - two
  - three

# Physical Science Chapter 6 Chemical Bonds Chapter Review

**Judd E. Hollander**



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**Chemistry: The Molecular Nature of Matter and Change** Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed ,2000

**Addison-Wesley Introduction to Physical Science** Michael B. Leyden, 1988 *ATI TEAS Strategies, Practice & Review with 2 Practice Tests* Kaplan Nursing, 2017-01-03 Provides comprehensive exam review as well as test taking strategies and study techniques *Discover Science: Teacher's annotated edition ,1991* Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today's

technological society **Sif: Chemistry 5na Tb** J. G. R. Briggs, 2009 *Superatoms* Puru Jena, Qiang Sun, 2021-12-01 Explore the theory and applications of superatomic clusters and cluster assembled materials Superatoms Principles Synthesis and Applications delivers an insightful and exciting exploration of an emerging subfield in cluster science superatomic clusters and cluster assembled materials The book presents discussions of the fundamentals of superatom chemistry and their application in catalysis energy materials science and biomedical sciences Readers will discover the foundational significance of superatoms in science and technology and learn how they can serve as the building blocks of tailored materials promising to usher in a new era in materials science The book covers topics as varied as the thermal and

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chemistry radiobiology tumor models and clinical data for targeted radionuclide therapy It outlines the current challenges and provides a glimpse at future directions With significant advances in cell biology and molecular engineering many targeting constructs are now available that will safely deliver these highly cytotoxic radionuclides in a targeted fashion A companion website includes the full text and an image bank      **Three-Dimensional Free-Radical Polymerization**

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