

- 10. The reaction is exothermic because the products have a lower enthalpy than the reactants.
- 11. The reaction is exothermic because the products have a lower enthalpy than the reactants.
- 12. The reaction is exothermic because the products have a lower enthalpy than the reactants.
- 13. The reaction is exothermic because the products have a lower enthalpy than the reactants.



Figure 10



Figure 11

Figure 10 shows the enthalpy change for the reaction of hydrogen and oxygen. The reaction is exothermic because the products have a lower enthalpy than the reactants.

Answer the questions in the margin.

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