



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Table of Contents Physical Science Grade 11 Term 3

1. Understanding the eBook Physical Science Grade 11 Term 3
 - The Rise of Digital Reading Physical Science Grade 11 Term 3
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade 11 Term 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade 11 Term 3
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade 11 Term 3
 - Personalized Recommendations
 - Physical Science Grade 11 Term 3 User Reviews and Ratings
 - Physical Science Grade 11 Term 3 and Bestseller Lists
5. Accessing Physical Science Grade 11 Term 3 Free and Paid eBooks
 - Physical Science Grade 11 Term 3 Public Domain eBooks
 - Physical Science Grade 11 Term 3 eBook Subscription Services
 - Physical Science Grade 11 Term 3 Budget-Friendly Options
6. Navigating Physical Science Grade 11 Term 3 eBook Formats

- ePub, PDF, MOBI, and More
 - Physical Science Grade 11 Term 3 Compatibility with Devices
 - Physical Science Grade 11 Term 3 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade 11 Term 3
 - Highlighting and Note-Taking Physical Science Grade 11 Term 3
 - Interactive Elements Physical Science Grade 11 Term 3
 8. Staying Engaged with Physical Science Grade 11 Term 3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade 11 Term 3
 9. Balancing eBooks and Physical Books Physical Science Grade 11 Term 3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade 11 Term 3
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science Grade 11 Term 3
 - Setting Reading Goals Physical Science Grade 11 Term 3
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science Grade 11 Term 3
 - Fact-Checking eBook Content of Physical Science Grade 11 Term 3
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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