

# Grade 11 Physical Science M...

records the change in the mass of magnesium as the reaction proceeds at regular intervals. The balanced chemical equation for the reaction is:



The change in the mass of magnesium during the reaction is shown on the graph below.



6.2.1 Identify the limiting agent in this reaction. Give a reason for the answer. (2)

6.2.2 Calculate the number of moles of unreacted hydrochloric acid in the test tube after 3 minutes. (7)

[19]

## Hide Answers

### QUESTION/VRAAG 6

6.1.1  $2\text{H}_2\text{O}_2\text{(aq)} \checkmark \rightarrow 2\text{H}_2\text{O(l)} + \text{O}_2\text{(g)} \checkmark$  (2)

6.1.2 The catalyst lowers the activation energy of the reaction ✓✓

Accept: catalyst speeds up the reaction

In Katalisator verlaag die aktiveringsenergie van die reaksie ✓✓

Aanvaar: katalisator laat die reaksie vinniger plaasvind (2)

6.1.3 **OPTION 1/OPSIE 1**

$$n = \frac{V}{V_m} \checkmark$$

$$n = \frac{0,6}{24,45} \checkmark$$

$$n = 0,0245 \text{ mole/mol O}_2 \text{ produced/gevorm}$$

$$\text{H}_2\text{O}_2 : \text{O}_2 \checkmark$$

$$2 : 1 \checkmark$$

$$n = 0,049 \text{ mole/mol H}_2\text{O}_2 \text{ reacted/reageer}$$

$$n = \frac{m}{M} \checkmark$$

$$0,049 = \frac{m}{34} \checkmark$$

**OPTION 2/OPSIE 2**

From the balanced equation:

Vanaf gebalanseerde vergelyking:



$$X = \frac{68 \times 0,6}{24,45} \checkmark$$

$$X = 1,67 \text{ g} \checkmark$$

# Physical Science Exemplar November 2014 Grade 11

**Thirumalaisamy P.  
Velavancorresponding**



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web sample leave application for tour by students and employees these sample leave application formats for tour on holiday with family within the country or foreign country both students and employees can follow these formats you can modify the format as

**how to write leave application in office for going to trip tour leave** - Jun 02 2022

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