#### Quiz 4.3 Chemical Reactions

Combination

## C. Types of Chemical Reactions

Directions: Classify the following unbalanced chemical equations according to the six types c chemical reaction.

Double displacement

Decomposition Single displacement	Combustion Acid-base
1	. NaOH + KNO3 → NaNO3 + KOH
	. CH <sub>4. +</sub> O <sub>2</sub> → CO <sub>2</sub> + H <sub>2</sub> O
3. 4. 5. 6. 7. 8. 9. 10 11 11 11	. <u>Fe +</u> NaBr → FeBr <sub>3</sub> + Na
	. CaSO <sub>4</sub> + Mg(OH) <sub>2</sub> → Ca(OH) <sub>2</sub> + MgSO <sub>4</sub>
	. NH <sub>4</sub> OH + HBr → H <sub>2</sub> O + NH <sub>4</sub> Br
	i. P <sub>4</sub> + O <sub>2</sub> → P <sub>2</sub> O <sub>5</sub>
	'. NaNO₃ → NaNO₂ <u>+ O</u> ₂
	. C <sub>18</sub> H <sub>18</sub> + O <sub>2</sub> → CO <sub>2</sub> + H <sub>2</sub> O
	. H <sub>2</sub> SO <sub>4</sub> + NaOH → Na <sub>2</sub> SO <sub>4</sub> + H <sub>2</sub> O
	0. NiSO4 + Li3PO4 → Ni3 (PO4) 2 + Li2SO4
	1. Fe + CuSO <sub>4</sub> → FeSO <sub>4</sub> + Cu
	2. Mg + O₂ → MgO
	3. C <sub>4</sub> H <sub>10</sub> + O <sub>2</sub> → CO <sub>2</sub> + H <sub>2</sub> O
	4. HC₂H₃O₂ + NaHCO₃ → NaC₂H₃O₂+ CO₂ + H₂O
	5. CuSO <sub>4</sub> + NaOH → Cu(OH) 2 + Na <sub>2</sub> SO <sub>4</sub>

# **Section Quiz Describing Chemical Reactions**

Northwestern University (Evanston, III.). Dental School

#### **Section Quiz Describing Chemical Reactions:**

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