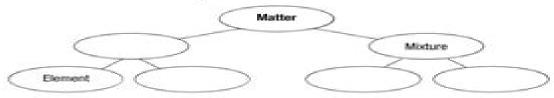
Chapter 2 Properties of Matter

Section 2.1 Classifying Matter (pages 38-44)

This section explains how materials are classified as pure substances or mixtures. It discusses types of pure substances and mixtures.

Reading Strategy (page 38)

Summarizing As you read, complete the classification of matter in the diagram below. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.



Pure Substances (page 39)

- 1. Is the following sentence true or false? Every sample of a pure substance has exactly the same composition and the same properties.
- 2. What are the two categories of pure substances?

and the second s	for
a-	D

Elements (pages 39-40)

- 3. What is an element?
- 4. Is the following sentence true or false? The smallest particle of an element is an atom. _____
- 5. Why does an element have a fixed, uniform composition? _____
- Circle the letter before each element that is a gas at room temperature.
 - a. carbon.

b. oxygen.

c. mercury

d. nitrogen

Match each element to its correct symbol.

Element	Symbol
 _ 7. aluminum	a. C
 _ 8. gold	b. A1
 _ 9. carbon	c. Au

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Table of Contents Section 3reptiles Answer Key

- 1. Understanding the eBook Section 3reptiles Answer Key
 - The Rise of Digital Reading Section 3reptiles Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Section 3reptiles Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 3reptiles Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Section 3reptiles Answer Key
 - Personalized Recommendations

- Section 3reptiles Answer Key User Reviews and Ratings
- Section 3reptiles Answer Key and Bestseller Lists
- 5. Accessing Section 3reptiles Answer Key Free and Paid eBooks
 - Section 3reptiles Answer Key Public Domain eBooks
 - Section 3reptiles Answer Key eBook Subscription Services
 - Section 3reptiles Answer Key Budget-Friendly Options
- 6. Navigating Section 3reptiles Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Section 3reptiles Answer Key Compatibility with Devices
 - Section 3reptiles Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 3reptiles Answer Key
 - Highlighting and Note-Taking Section 3reptiles Answer Key
 - Interactive Elements Section 3reptiles Answer Key
- 8. Staying Engaged with Section 3reptiles Answer Key
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 3reptiles Answer Key
- 9. Balancing eBooks and Physical Books Section 3reptiles Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 3reptiles Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 3reptiles Answer Key
 - $\circ\,$ Setting Reading Goals Section 3reptiles Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 3reptiles Answer Key
 - Fact-Checking eBook Content of Section 3reptiles Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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