

Factors Affecting the Rate of a Chemical Reaction

The following events must occur before a reaction can proceed:

1. The reactant particles must collide with each other.
2. The collisions must be of enough energy to overcome the 'energy barrier', called the activation energy (more about this on the next slide).
3. The reactants must form new bonds to produce products.

This presentation is useful as a refresher for those of you starting Unit 2 of the A-level chemistry course.

Section 4 Reinforcement Chemical Reaction Rates And Energy

Xiang Xie

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