

CHAPTER 13

Rates of Reaction

■ SOLUTIONS TO EXERCISES

Acetaminophen is a common pain reliever. Acetaminophen reacts with bromine water to form a colorless solution. The color change is used to determine the concentration of acetaminophen in a solution. Acetaminophen reacts with bromine water to form a colorless solution. The color change is used to determine the concentration of acetaminophen in a solution.

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- 13.5. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$.

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